

Lecture 4b: Modelling of Distributed Parameter Systems

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(Plus material from Ian Cameron)

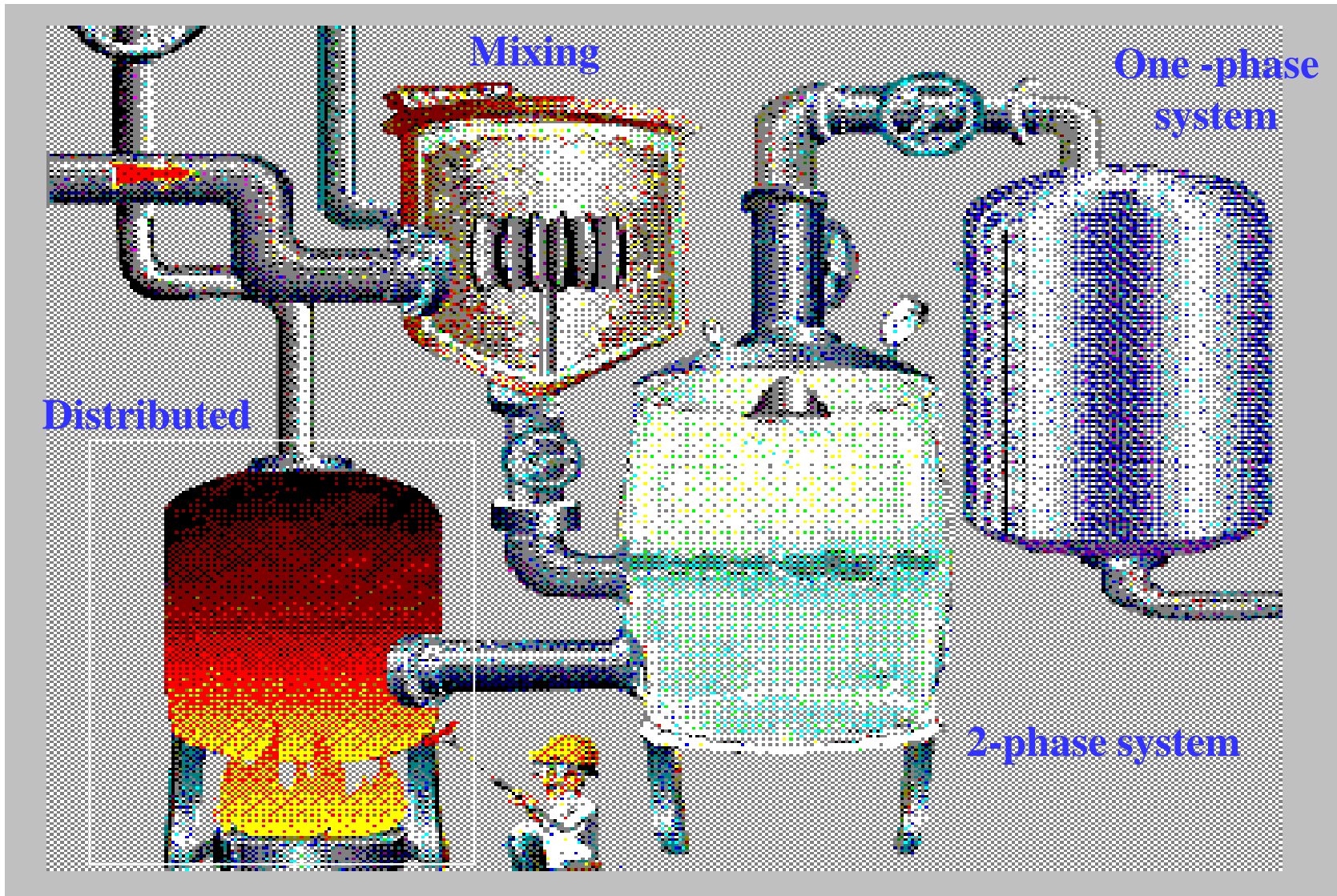
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Overview of lecture 4b

- ❖ Origin of DPS models
- ❖ Modelling of DPS
 - Balance volumes
 - Conservation balances
 - Boundary and initial conditions
- ❖ Classification of DPS model equations
- ❖ Lumped parameter models for DPS
- ❖ Examples

The Origin of DPS Models



The Origin of DPS Models: Differential form of conservation balances

For a conserved extensive quantity Φ and its related potential φ

Flow terms:

❖ convective flows: $J_C = v \Phi$

❖ diffusive flows: $J_D = -D \text{grad } \varphi$
(assume D =constant and no cross-effect)

Co-ordinate system independent form:

$$\frac{\partial \hat{\Phi}}{\partial t} = D(\nabla^2 \varphi(r, t)) - \nabla \bullet (\hat{\Phi}(r, t)v(r, t)) + \hat{q}(r, t)$$

Conservation in rectangular co-ordinates

Differential operator, rectangular co-ordinate system

$$\frac{\partial \hat{\Phi}}{\partial t} = D \left(\frac{\partial^2 \varphi}{\partial x^2} + \frac{\partial^2 \varphi}{\partial y^2} + \frac{\partial^2 \varphi}{\partial z^2} \right) - \left(\frac{\partial \hat{\Phi}}{\partial x} v_x + \frac{\partial \hat{\Phi}}{\partial y} v_y + \frac{\partial \hat{\Phi}}{\partial z} v_z \right) + \hat{q}$$

“dynamic”

“diffusion”

“convection”

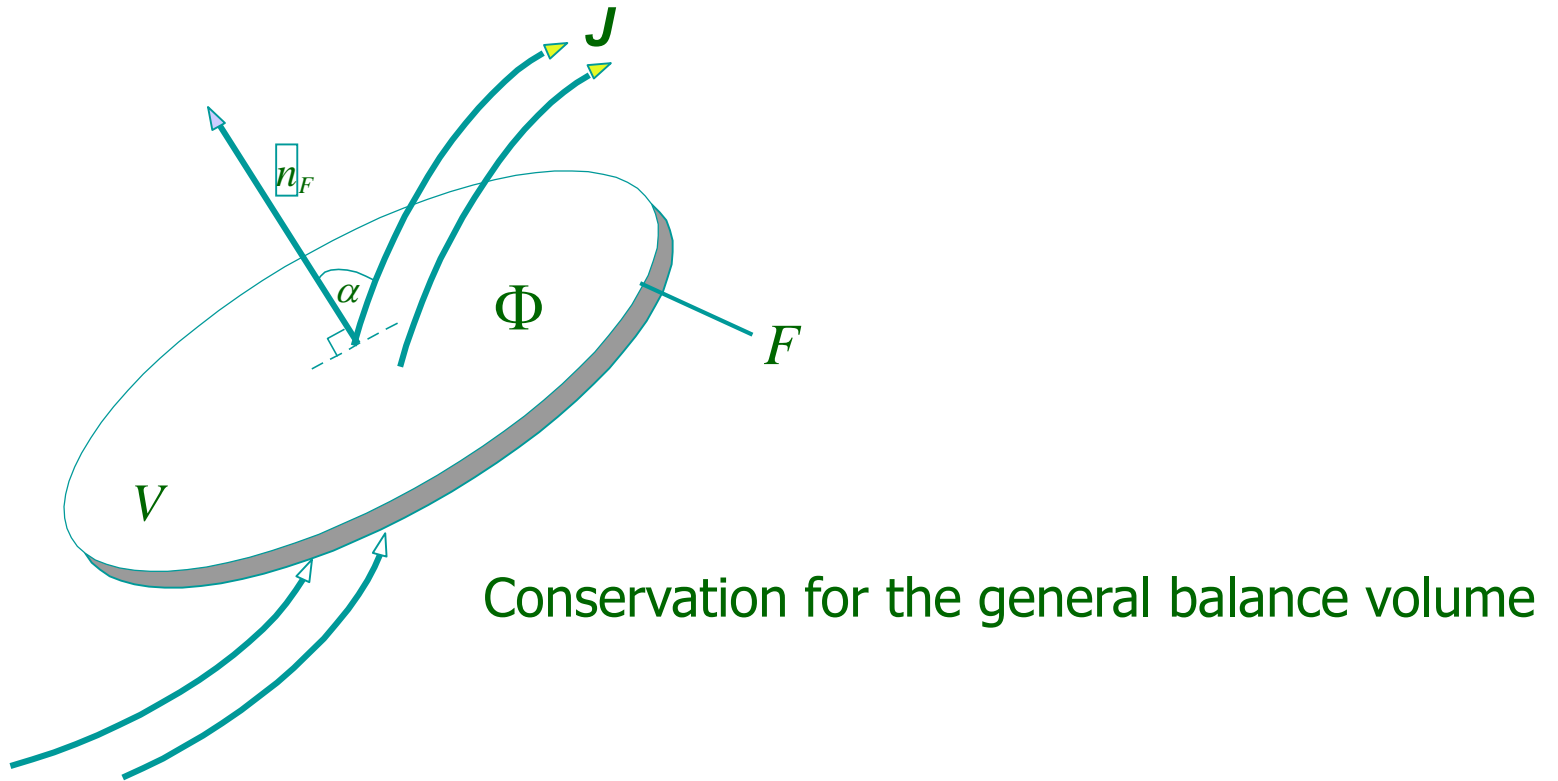
“source”

- Parabolic partial differential equation
- Induced algebraic equations
 - extensive-intensive relations
 - rate equations (transfer and reaction)

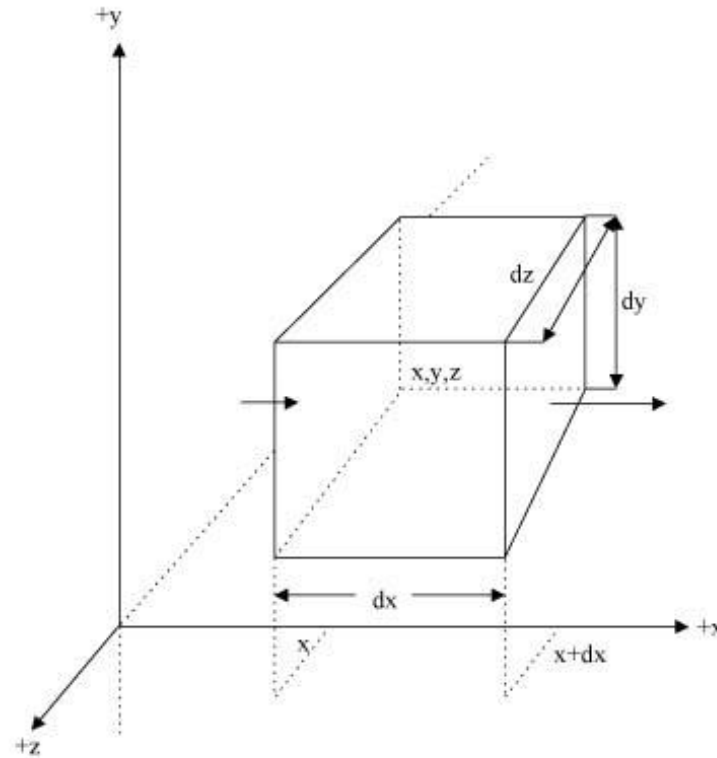
Balance volumes for DPSs

- ❖ “Distributed” balance volumes
 - uniform phase
 - uniform flow pattern
- ❖ Size of the balance volume
- ❖ Shape of the balance volumes
 - co-ordinate system
 - rectangular
 - cylindrical
 - spherical

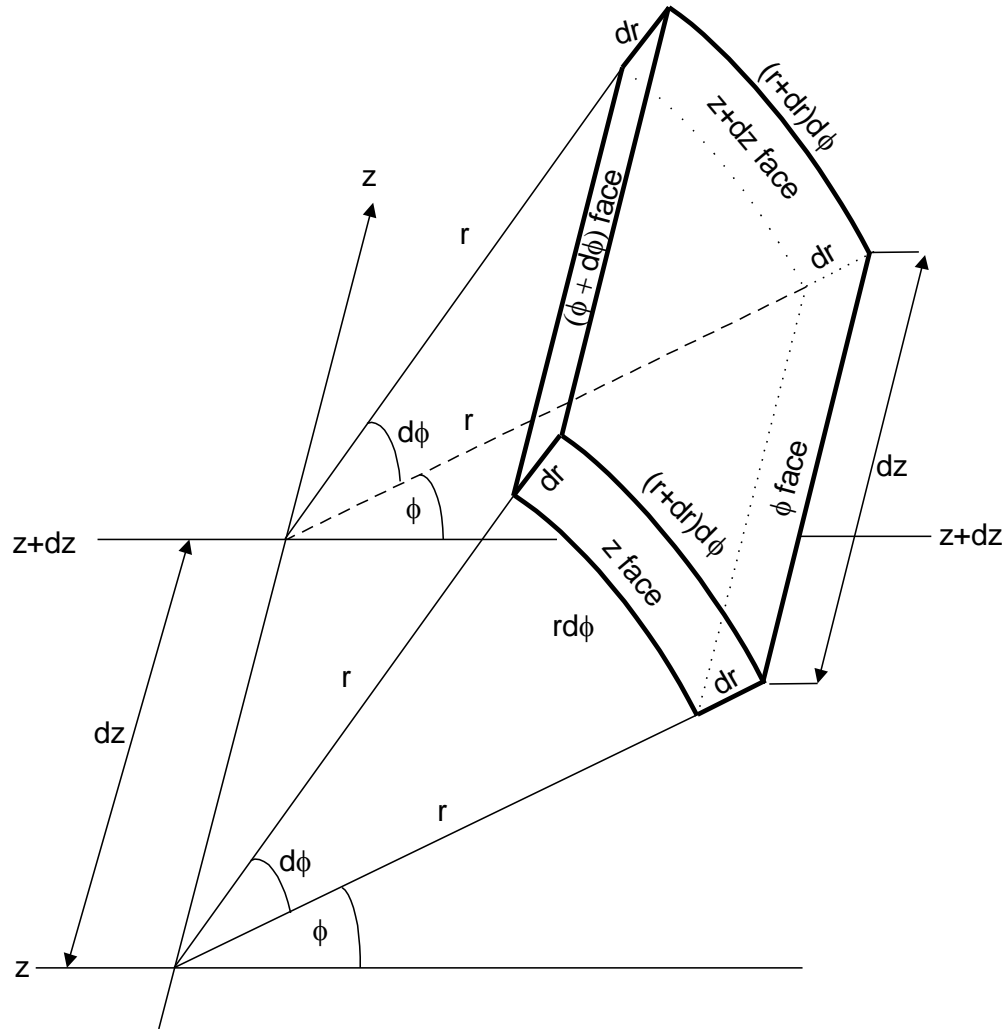
Balance or “control” volume



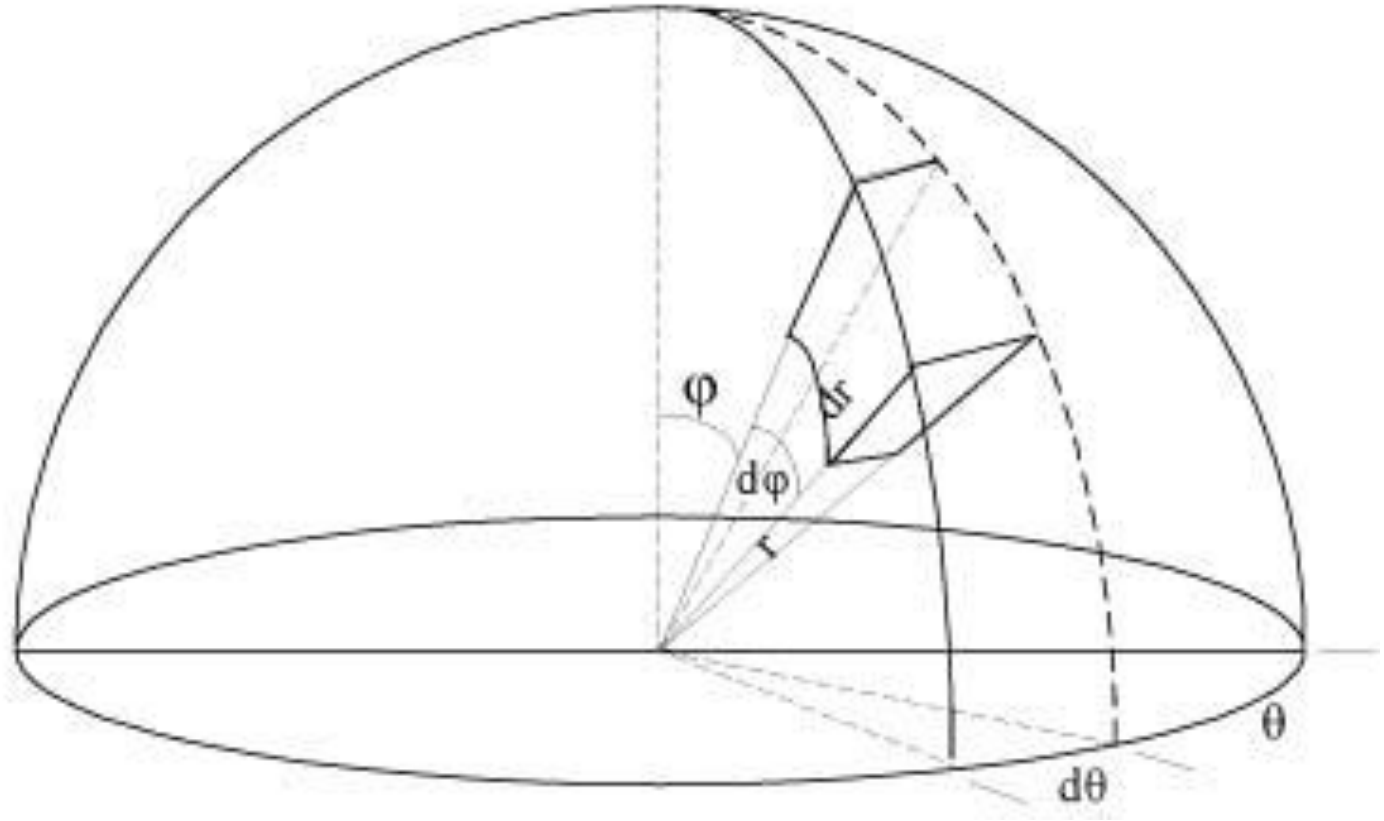
Balance Volume in rectangular co-ordinates



Balance volume in cylindrical co-ordinates



Balance volume in spherical co-ordinates



Derivation of DPS models – using microscopic balances

- ❖ Use of an arbitrary finite volume.
- ❖ Consider volume reduced to a point.
- ❖ Applicable to all geometries.
- ❖ Transformation to other co-ordinate systems is possible.

Mass balance in rectangular co-ordinates

Balance volume: $dV = dx.dy.dz$

Mass conservation within dV :

$$\frac{\partial(M)}{\partial t} = \frac{\partial(\rho.dx.dy.dz)}{\partial t} = (\rho v_x - \rho v_{x+dx}).dy.dz + (\rho v_y - \rho v_{y+dy}).dx.dz + (\rho v_z - \rho v_{z+dz}).dx.dy$$

x-direction

y-direction

z-direction

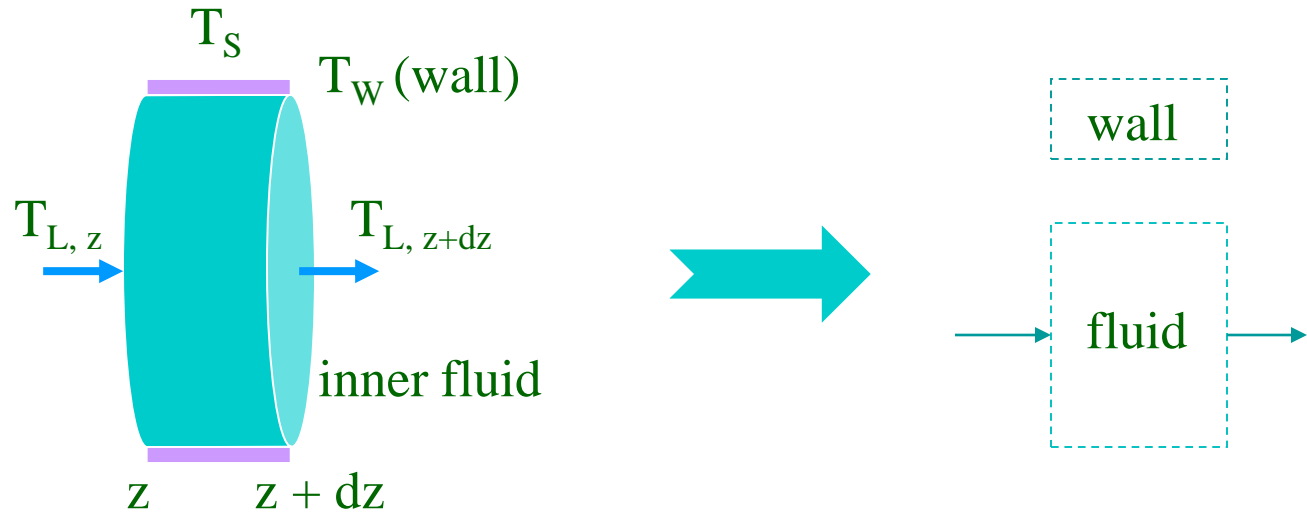
Divide by $dx.dy.dz$:

$$\frac{\partial(M)}{dV\partial t} = \frac{\partial\rho}{\partial t} = \frac{(\rho v_x - \rho v_{x+dx})}{dx} + \frac{(\rho v_y - \rho v_{y+dy})}{dy} + \frac{(\rho v_z - \rho v_{z+dz})}{dz}$$

In the limit:

$$\frac{\partial\rho}{\partial t} = -\frac{\partial}{\partial x}\rho v_x - \frac{\partial}{\partial y}\rho v_y - \frac{\partial}{\partial z}\rho v_z = -\nabla.\rho v$$

Double pipe heat exchanger



Fluid energy conservation:

$$\frac{\partial E_L}{\partial t} = \frac{\partial (A_f \rho c_p T_L dz)}{\partial t} = F \hat{h}_{L,z} - F \hat{h}_{L,z+dz} + Q_{WL} = u \rho_L c_{PL} (T_z - T_{z+dz}) + U_L A_L dz (T_W - T_L)$$

In the limit:
$$\rho_L A_f c_L \frac{\partial T_L}{\partial t} = -u \rho_L A_f c_L \frac{\partial T_L}{\partial z} + U_L A_L (T_W - T_L)$$

Double pipe heat exchanger (cont.)

Wall energy conservation:

$$\frac{\partial E_W}{\partial t} = \frac{\partial(M_W c_{pw} T_W dz)}{\partial t} = Q_{SW} - Q_{WL} = h_{SW} A_S dz (T_S - T_W) + h_{WL} A_L dz (T_W - T_L)$$

In the limit:

$$M_W c_{pw} \frac{\partial(T_W)}{\partial t} = h_{SW} A_S (T_S - T_W) + h_{WL} A_L (T_W - T_L)$$

Final:

$$\frac{\partial T_L}{\partial t} = -u \frac{\partial T_L}{\partial z} + \frac{1}{\tau_L} (T_W - T_L)$$

$$\frac{\partial T_W}{\partial t} = \frac{1}{\tau_{SW}} (T_S - T_W) - \frac{1}{\tau_{WL}} (T_W - T_L)$$

BCs & ICs

$$T_L(z,0) = f_1(z) ; T_W(z,0) = f_2(z)$$

$$T_L(0,t) = T^{(i)}(t) \quad t \geq 0$$

$$T_W(0,t) = T^w(t) \quad t \geq 0$$

Packed-bed catalytic reactor

Assumptions (balances and constitutive equations)

- ❖ plug flow
- ❖ first order $A \rightarrow B$ reaction
- ❖ liquid bulk phase
- ❖ solid phase catalyst
- ❖ uniform in cross-section
- ❖ constant physico-chemical properties
- ❖ bulk temperature constant

Packed-bed catalytic reactor

Model equations

Component mass
$$\frac{\partial C_A}{\partial t} = D \frac{\partial^2 C_A}{\partial x^2} - F \frac{\partial C_A}{\partial x} - r_A$$

Energy
$$\frac{\partial \hat{U}}{\partial t} = \kappa \frac{\partial^2 \hat{U}}{\partial x^2} - F \frac{\partial \hat{U}}{\partial x} - \Delta H r_A - q_{tr}$$

Constitutive equations
$$\hat{U} = c_P \rho T \quad r_A = k_0 e^{-\frac{E}{RT}} C_A$$

$$q_{tr} = K(T - T_w)$$

Spherical catalyst pellet

Assumptions (balances & constitutive equations)

- ❖ overall mass and volume is constant
- ❖ first order $A \rightarrow B$ reaction
- ❖ no convection
- ❖ uniform in all directions
- ❖ constant physico-chemical properties

Spherical catalyst pellet

Model equations in spherical co-ordinate system

Component mass $\frac{\partial c}{\partial t} = \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial c}{\partial r} \right) - k_0 c e^{\frac{E}{RT}}$

Energy $\frac{\partial T}{\partial t} = \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial T}{\partial r} \right) - k_0 c e^{\frac{E}{RT}} \Delta H_R$

Note: Constitutive equations are substituted

Initial conditions

Set the values of the states at the initial time ($t = 0$)

Given as a modelling assumption

Examples:

$$T(x, 0) = f_1(x) \quad (f_1(x) \text{ is given})$$

$$c_A(x, 0) = c_0 \quad (c_0 \text{ is given})$$

Boundary conditions

❖ Relevant assumptions

- ◆ conditions on the boundaries
- ◆ balance volume shape (coordinate system)
- ◆ balance volume size (infinite)

❖ Number of independent boundary conditions

- ◆ along a co-ordinate direction
- ◆ equal to the order of partial derivatives

Boundary condition types

Dirichlet (1st type) condition

value set on boundary: $c_A(0, t) = c^*$

Neumann (2nd type) condition $\frac{\partial c_A}{\partial x}(0, t) = 0$
 flux set on boundary

Robbins (3rd type) condition

convective transfer $\frac{\partial c_A}{\partial x}(x_M, t) = K (c^* - c_A(x_M, t))$

Packed-bed catalytic reactor

Additional assumptions

- ❖ initial distribution uniformly constant
- ❖ “very long” reactor

Initial condition $C_A(x, 0) = C^* , T(x, 0) = T^*$

Boundary conditions

$$\frac{\partial C_A}{\partial x}(L, 0) = 0 , C_A(0, t) = C_A^{(i)}$$

$$\frac{\partial T}{\partial x}(L, 0) = 0 , T(0, t) = T^{(i)}$$

Spherical catalyst pellet

Additional assumptions

- ❖ given initial conditions
- ❖ heat and mass transfer on the surface

Boundary conditions

at the centre $\frac{\partial c}{\partial x} = \frac{\partial T}{\partial x} = 0$

at the surface

$$-\frac{\partial T}{\partial x} = \frac{Nu}{2}(T - g_1(t))$$

$$-\frac{\partial c}{\partial x} = \frac{Sh}{2}(c - g_2(t))$$

Initial conditions $T(x,0) = h_1(x), c(x,0) = h_2(x)$

Classification of DPS models

Partial differential part of the conservation equation

$$\frac{\partial \hat{\Phi}}{\partial t} = D \left(\frac{\partial^2 \varphi}{\partial x^2} + \frac{\partial^2 \varphi}{\partial y^2} + \frac{\partial^2 \varphi}{\partial z^2} \right) - \left(\frac{\partial \hat{\Phi}}{\partial x} v_x + \frac{\partial \hat{\Phi}}{\partial y} v_y + \frac{\partial \hat{\Phi}}{\partial z} v_z \right) + \hat{q}$$

Associated algebraic equation

$$t = D x^2 + D y^2 + D z^2 - v_x x - v_y y - v_z z$$

Geometry of the 2nd order curve

- parabola: $D \neq 0$
- hyperbola (degenerate): $D=0, v \neq 0$
- ellipse: steady-state, $D \neq 0$

Lumped parameter models for DPSs

Conceptual steps in **lumping**:

- ◆ divide balance volume into sub-volumes
- ◆ lump each sub-volume using perfect mixing
- ◆ convection → in- and out-flows of lumps
- ◆ diffusion → multidirectional in- and out-flows
- ◆ same sources for every lump
- ◆ balance equations for every lump
- ◆ boundary conditions → in- and outflows of lumps on system boundary

DPS model of a double-pipe heat exchanger

Model equations

Variables $0 \leq z \leq L$, $0 \leq t$, $T_L(z, t)$, $T_w(x, t)$

Energy balances $\rho_L A_L c_f \frac{\partial T_L}{\partial t} = -u \rho_L A_f c_L \frac{\partial T_L}{\partial z} + h_L A_L (T_w - T_L)$

$$M_w c_w \frac{\partial T_w}{\partial t} = h_s A_s (T_s - T_w) - h_L A_L (T_w - T_L)$$

Initial conditions $T_L(z, 0) = f_1(z)$

$$T_w(z, 0) = f_2(z)$$

Boundary condition $T_L(0, t) = T^{(i)}(t)$ for all $t \geq 0$.

Lumped model of a double-pipe heat exchanger

Model equations

Variables $(T_L^{(k)}(t), T_w^{(k)}(t), k = 1, 2, 3), 0 \leq t$

Energy balances
$$\frac{dT_L^{(k)}}{dt} = 3u(T_L^{(k-1)} - T_L^{(k)}) + \frac{1}{\tau_L}(T_w^{(k)} - T_L^{(k)})$$

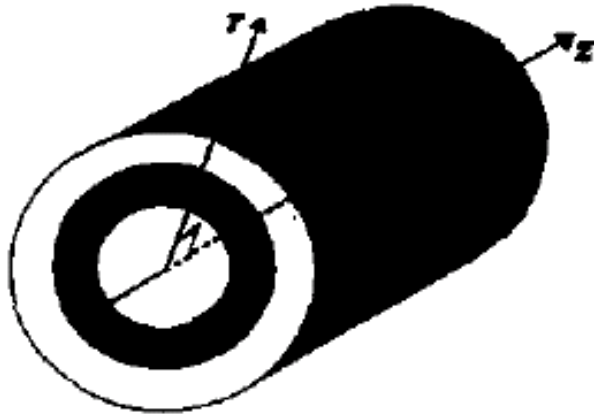
$$k = 1, 2, 3, \quad T_L^{(0)}(t) = T_L^{(i)}(t)$$

$$\frac{dT_w^{(k)}}{dt} = \frac{1}{\tau_{sw}}(T_s - T_w^{(k)}) - \frac{1}{\tau_{wL}}(T_w^{(k)} - T_L^{(k)})$$

$k = 1, 2, 3$

Initial conditions $T_L^{(k)}(0) = f_1^{(k)}, k = 1, 2, 3$
 $T_w^{(k)}(0) = f_2^{(k)}, k = 1, 2, 3$

Modelling exercise – 4b: DPS model for a cylindrical catalyst (**Problem II – home exercise**)



Step 1: Identify the phenomena to be considered:

- Conduction of mass and energy
- Reaction
- Single phase
- No accumulation

Step 2: Retrieve the general form of the balance volume equations:

- Select coordinate system
- Write conservation equation (in derivative form) for the corresponding coordinate system

Step 3: Generate final form of the model:

- Remove the terms not needed
- Retrieve from library the needed terms
- Add constitutive models
- Add initial and boundary conditions

$$\frac{\partial \hat{\Phi}}{\partial t} = D \left(\frac{1}{r} \frac{\partial}{\partial r} \left(\frac{\partial f}{\partial r} \right) + \frac{1}{r^2} \frac{\partial^2 f}{\partial \varphi^2} + \frac{\partial^2 f}{\partial z^2} \right) - \left(\left(\frac{\partial \hat{\Phi}}{\partial r} \right) \hat{r} + \left(\frac{1}{r} \frac{\partial \hat{\Phi}}{\partial \varphi} \right) \hat{\varphi} + \frac{\partial \hat{\Phi}}{\partial z} \hat{z} \right) + \hat{q}$$