

New Vistas in Chemical Product & Process Design

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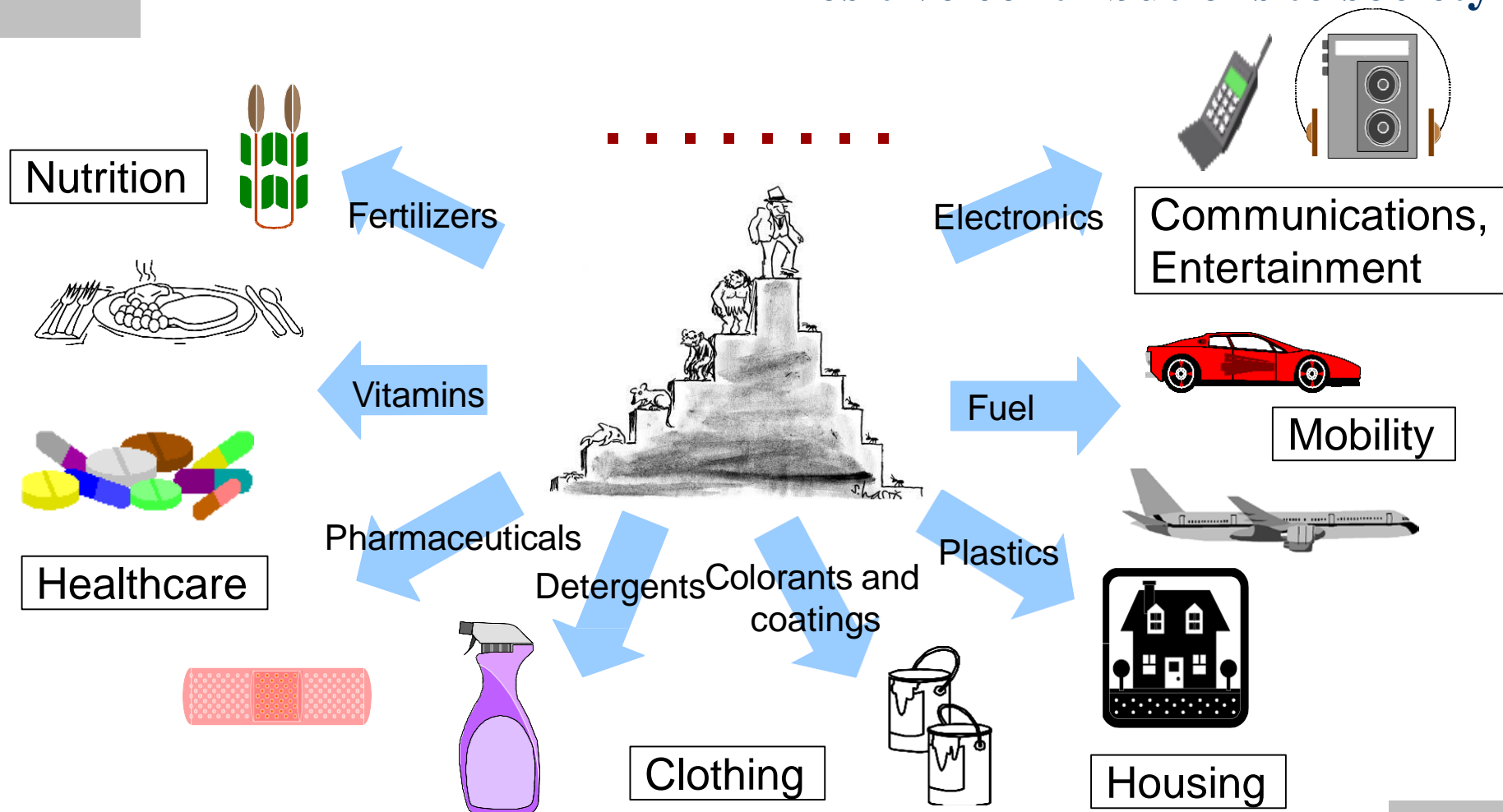
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The big picture

SPEED Master of the planet earth – how did we get there?

Positive contributions to society



Our survival depends on the products we make from the resources we have

The chemical product tree

Question of what, why & when (how)?

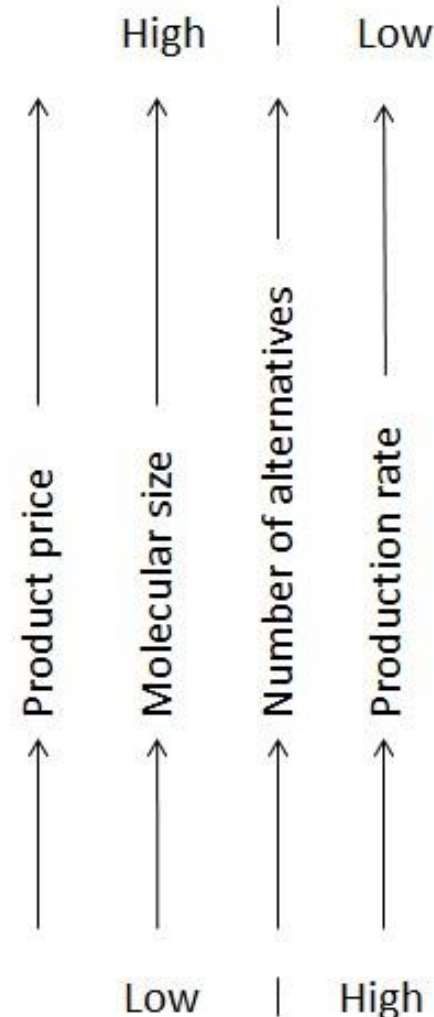
Refined chemicals & Consumer products (~3000)
Plastics, Pharmaceuticals, Dyes, Solvents, Fertilizers, Fibres, Dispensers, Cosmetics...



Intermediate Products (~300)
Methanol, Vinyl chloride, Styrene, Urea, Formaldehyde, Ethylene oxide, Acetic acid, Acrylonitrile, Cyclohexane, Acrylic acid,...

Basic Products (~20)
Ethylene, Propylene, Butadiene, Benzene, Synthesis-gas, Acetylene, Ammonia, Sulfuric acid, Sodium hydroxide, chlorine, ...

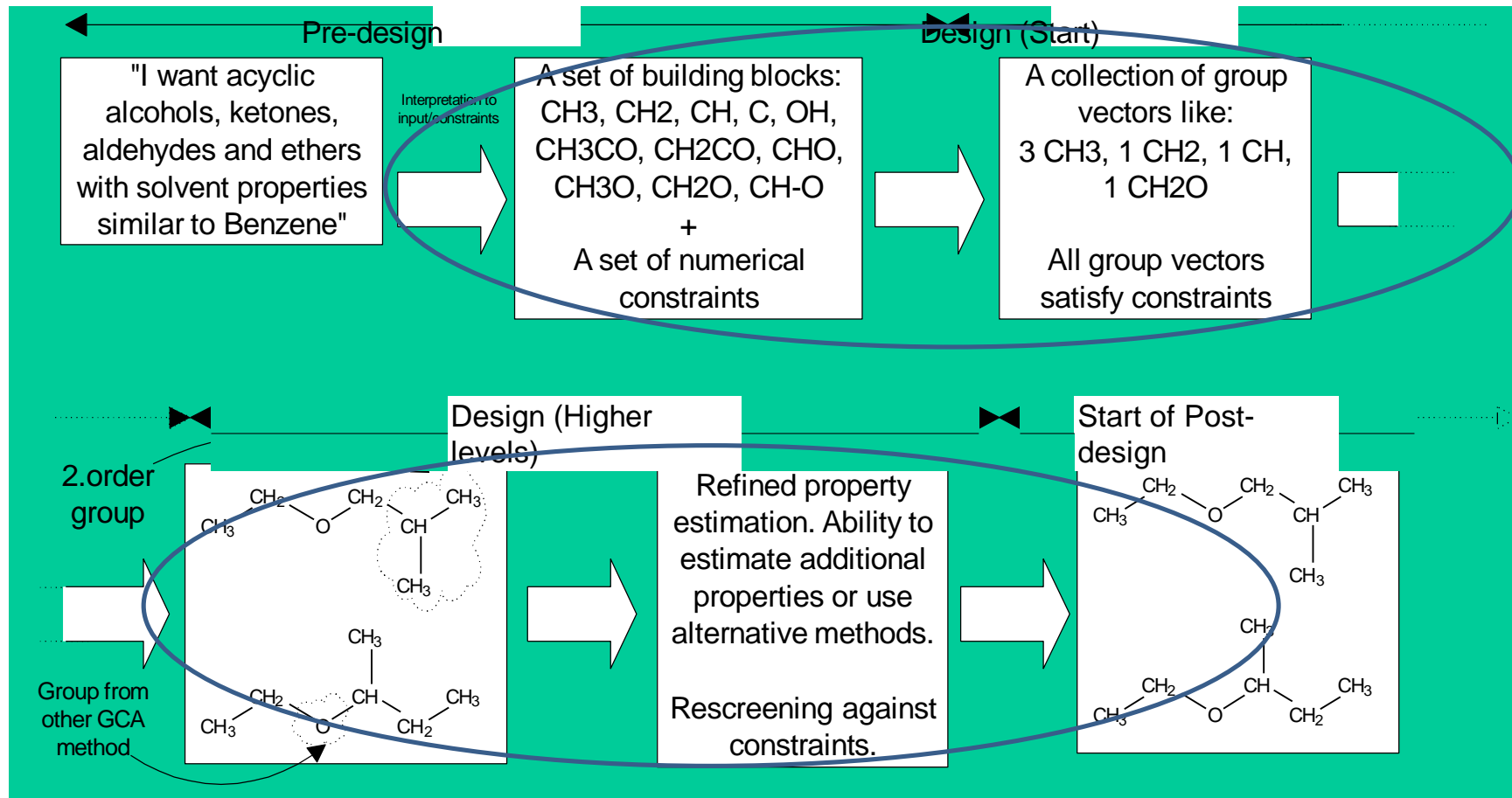
Raw Materials (~10)
Petroleum, Natural Gas, Biomass, Rock, Salt, Phosphate, Sulfur, Air, Water, ...



| Product Type | Single species | | Multiple species | | Devices |
|------------------------|---|------------------------------------|---|--|---|
| | Small | Large | Formulated | Functional | |
| Issues | | | | | |
| Examples | Solvents, refrigerants | APIs, surfactants, membranes | Blended fuels, solvents, lubricants | Detergents, personal care, healthcare, medicinal | Fuel-cells, microcapsules, hemodialysis device |
| Key factors in design | Cost of production | | Cost and product functions | Product functions | |
| | N.A. | Time to market (speed) | | | |
| Basis of design | Process flowsheet (unit operations) | Synthesis routes (chemistry) | Blend properties and stability | Microstructure | Device form and constituent materials |
| Measure of performance | Closely related to process | Closely related to application | Closely related to application | Consumer satisfaction | Consumer satisfaction |
| Challenges in design | Translation of needs to molecular structure | Translation of needs to properties | Delivering product functions defined by needs | | Translation of needs to product material properties and configuration |
| | Large number of product alternatives | | Trial & error approach (non-optimal product) | | |
| Risks | Feedstock availability | | | Absence of engineering science knowledge | |
| | Environmental impact and sustainability | | | | |

The concepts

Find molecules with desired properties (also valid for mixtures)



Computer aided molecular design (generate & test)

Generation of Alternatives

Groups as building blocks: CH_3 -, - CH_2 -, - OH , CH_3CO -, ... (a set of about 180 groups available)

Structural constraints (acyclic molecules)

$$\sum_{j=1}^m (2 - u_j) n_j = 2q \quad \sum_{i \neq j}^m n_i = n_j (u_j - 2) + 2 \quad \forall j$$

Size constraints

$$n_j^l \leq n_j \leq n_j^u \quad \forall j \quad 2 \leq \sum_{j=1}^m n_j \leq n_{\max} \quad n = \sum_{j=1}^m n_j$$

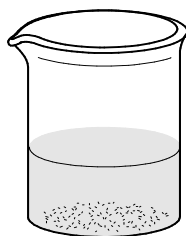
Example: $\text{CH}_3 - \text{CH}_2 - \text{OH}$: Ethanol (3 groups)

Evaluation of Generated Molecules: Models

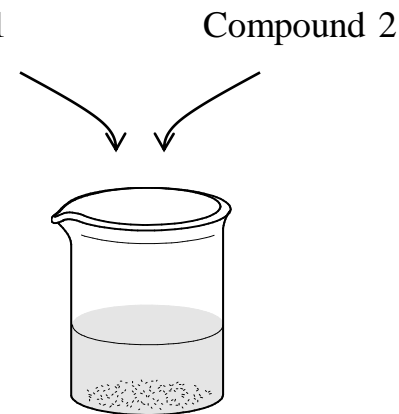
Group contribution (GC) methods for property prediction

GC MODELS

Pure compounds



Mixtures



Examples:

Examples:

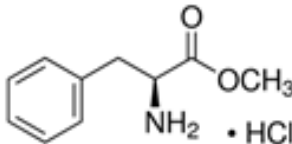
UNIFAC (VLE, LLE, SLE)

$$V_c = 17.5 + \sum_j n_j \Delta_j \quad \text{Joback \& Reid 1987}$$

$$\ln \gamma_i = \ln \gamma_i^{COM} + \ln \gamma_i^{RES}$$

$$V_c - V_{c0} = \sum_i N_i V_{c1i} + w \sum_j M_j V_{c2j} + z \sum_k O_k V_{c3k}$$

Gani, et al. 2001, 2007, 2012, 2013, 2014

| | | |
|---|--|---------------------------|
| Compound: L-Phenylalanine methyl ester hydrochloride Molecular formula: C6H5CH2CH(NH2)COOCH3 · HCl | Molecular structure  | |
| First-order groups | Occurrences | Group-contribution |
| CH3 | 1 | 0.7473 |
| aCH | 5 | 0.4297 |
| aC-CH2 | 1 | -0.2869 |
| COO (except as above) | 1 | 1.966 |
| -HCl | 1 | 0.4662 |
| Second-order groups | Occurrences | Group-contribution |
| -- | -- | -- |
| Third-order groups | Occurrences | Group-contribution |
| CH(NH3+Cl-)-(CH2) <i>n</i> -COO- (CH2) <i>m</i> <i>n</i> in (0,...,2); <i>m</i> in (0,...,2) | 1 | -0.9282 |
| $\exp\left(\frac{T_m}{T_{m0}+a \cdot MW}\right) = \sum_i N_i C_i + \sum_j M_j D_j + \sum_l E_l O_k \Rightarrow T_m^{\text{pred}} = 430.9 \text{ (Note: } T_{m0} = 237.0611 \text{ K; } a = -0.0914 \text{ K mol g}^{-1}\text{). The experimental value of } T_m \text{ is } 433.2 \text{ K. Hence, absolute deviation} = 2.2 \text{ K}$ | | |

Polymer Design: MINLP Applied to CAMD

$$F_{obj} = \sum [(P_i - P_i^*)/P_i^*]^2$$

s.t.

$$353 \text{ K} < T_g = [\sum (Y_k n_k)] / [\sum (M_k n_k)] < 393 \text{ K}$$

$$1.4 < \rho = [\sum (M_k n_k)] / [\sum (V_k n_k)] < 1.5 \text{ g/cm}^3$$

$$0.0045 < W = [\sum (H_k n_k)] / [\sum (M_k n_k)] < 0.0055 \text{ g H}_2\text{O/g polymer}$$

$$2 \leq \sum y_j \leq 3 \quad ; \quad y_j : 0 \text{ or } 1 \text{ for } j=1,7$$

$$2 \leq \sum y_k n_k \leq 9$$

-CH₂- ; -CO- ; -COO- ; -O- ; -CONH- ; -CHOH- ; -CHCL-

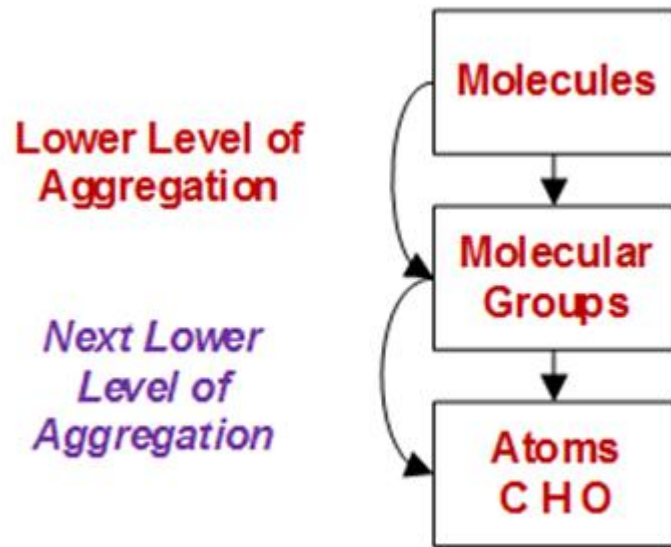
Note: objective function and constraints are non linear; n_k , the optimization variables are integer (0-9)

Polymer repeat units designed for bottle-stops; synthetic fibers; coatings, etc (CACE 2009)

Extension to process flowsheet synthesis

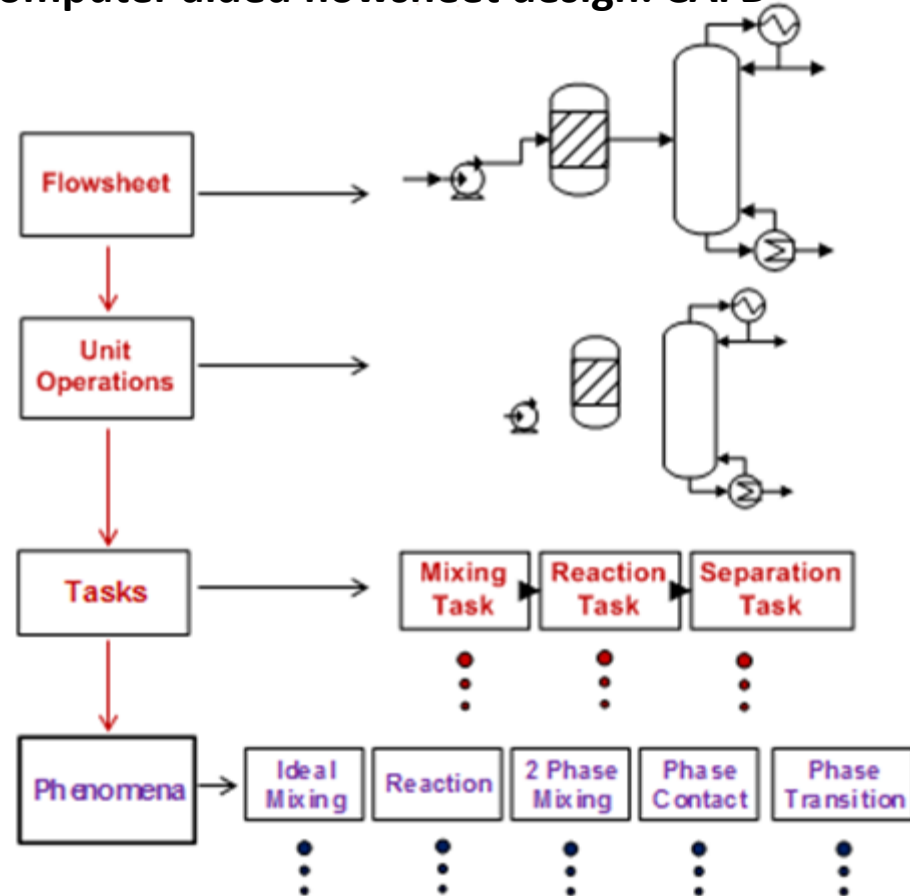
CAMD versus CAFD

Computer aided molecular design: CAMD



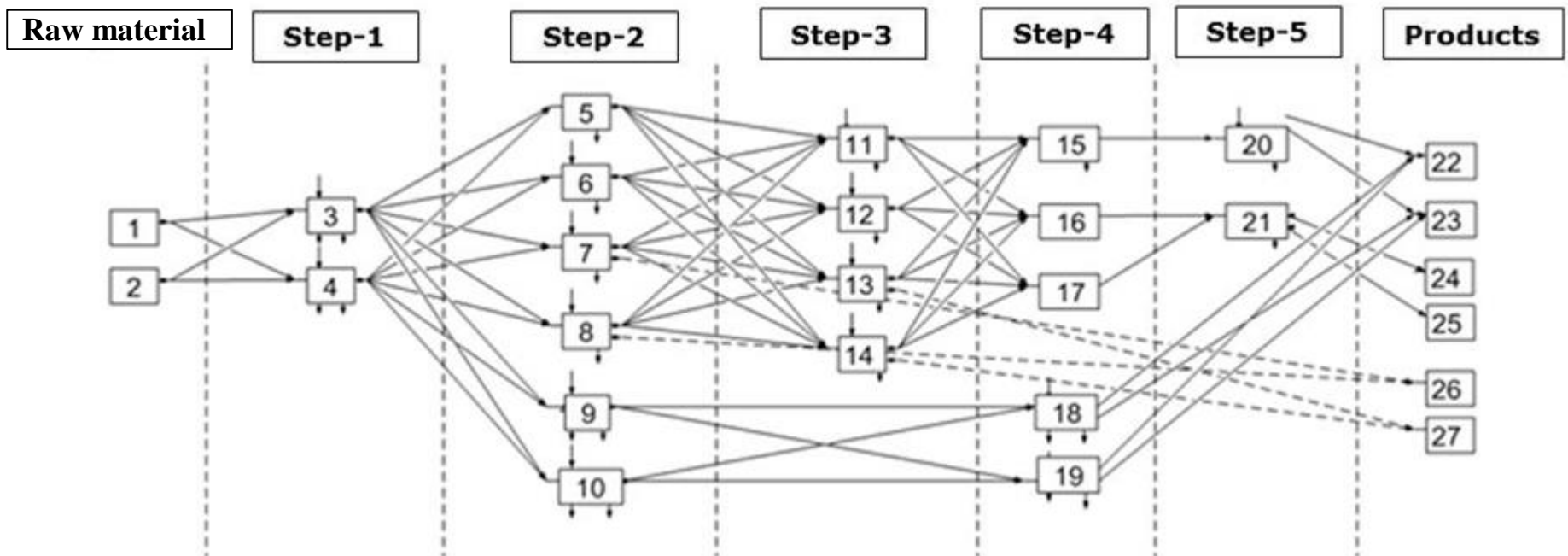
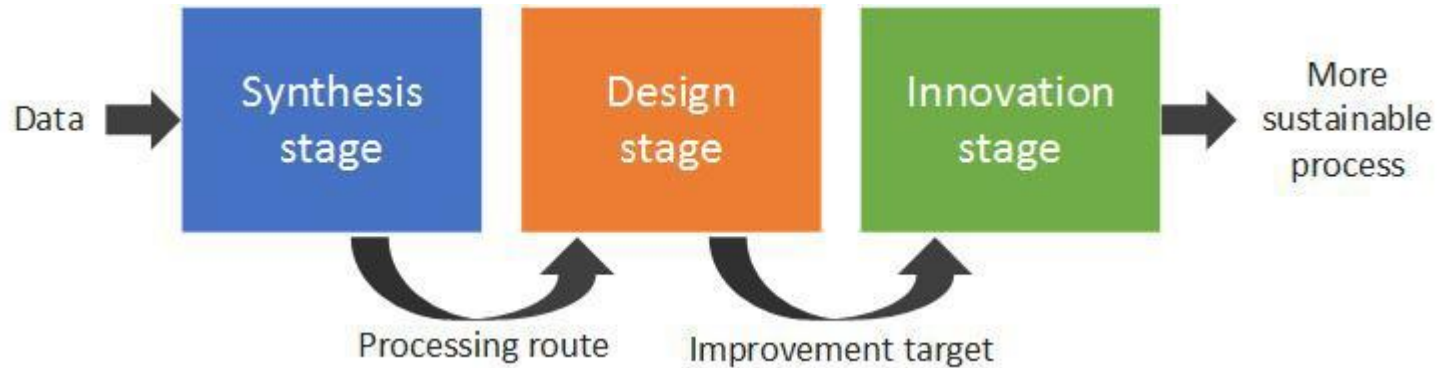
Gani et al. (1994)

Computer aided flowsheet design: CAFD



Babi et al. (2014)

Sustainable Product-Process Development

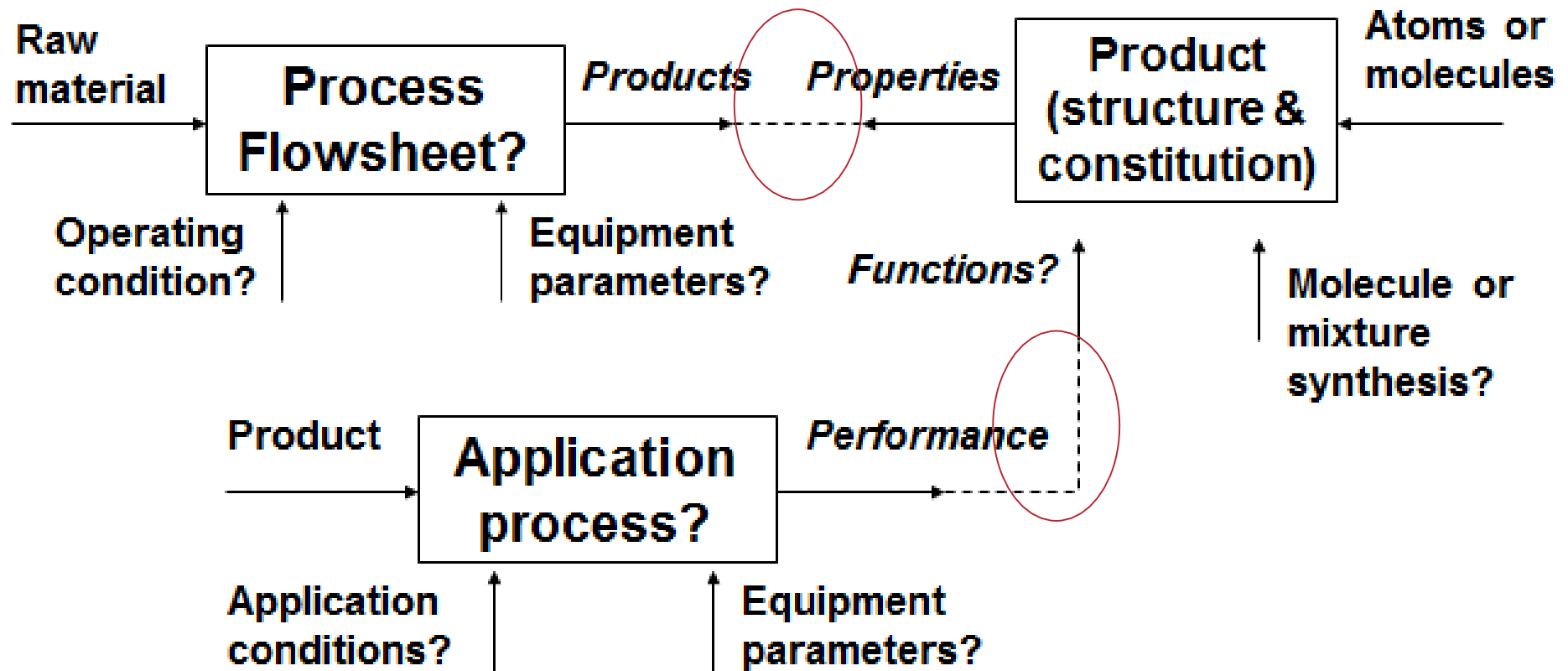


Applications: Biorefinery; CCU; CAMD; Waste-water treatment;

The problem definitions

a: Process Design

b: Product Design



c: Product Application

Problems may have multiscale & multidiscipline features

SPEED Mathematical generic problem formulation

$$\text{Fobj} = \min \{C^T \underline{y} + f(\underline{x}, \underline{y}, \underline{u}, \underline{d}, \underline{\theta}) + S_e + S_i + S_s + H_c + H_p\} \quad (1)$$

$$0 = h_1(\underline{x}, \underline{y}) \quad \text{process constraints (Eq. 2)}$$

$$0 = P(f, \underline{x}, \underline{y}, \underline{d}, \underline{u}, \underline{\theta}) \quad \text{process model (Eq. 3)}$$

$$\underline{\theta} = \underline{\theta}(f, \underline{x}, \underline{y}) \quad \text{product-property model (Eq. 4)}$$

$$l_1 \leq g_1(\underline{x}, \underline{u}, \underline{d}) \leq u_1 \quad \text{process variable constraints (Eq. 5)}$$

$$l_2 \leq g_2(\underline{x}, \underline{y}) \leq u_2 \quad \text{molecular structure constraints (Eq. 6)}$$

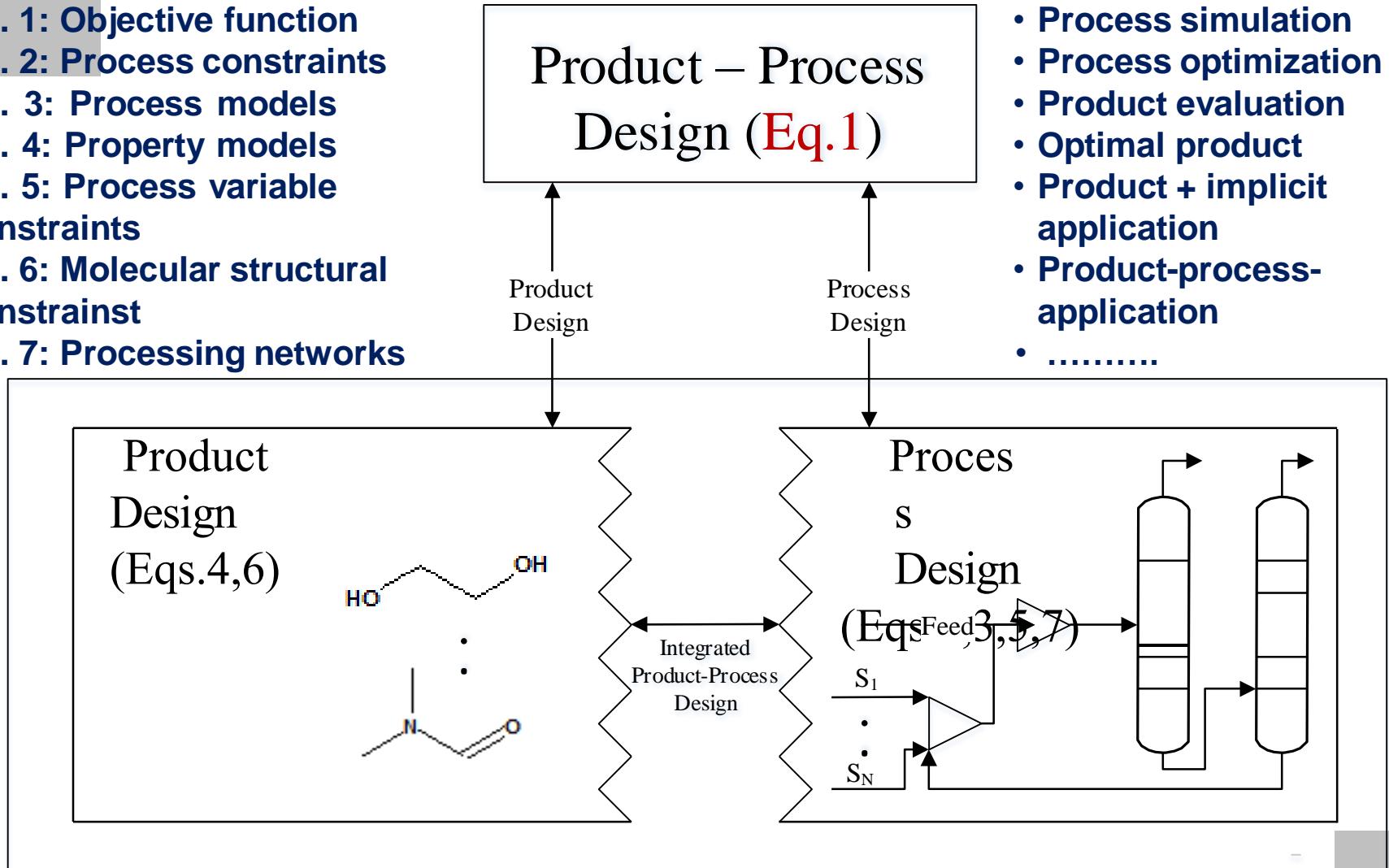
$$B \underline{x} + C^T \underline{y} \geq D \quad \text{process networks (Eq. 7)}$$

\underline{x} : real-process variables; \underline{y} integer-decision variables;
 \underline{u} : process design variables; \underline{d} : process input variables;
 $\underline{\theta}$: property; B, C, D coefficient matrices

SPEED Different problem formulations

- Eq. 1: Objective function
- Eq. 2: Process constraints
- Eq. 3: Process models
- Eq. 4: Property models
- Eq. 5: Process variable constraints
- Eq. 6: Molecular structural constraint
- Eq. 7: Processing networks

- Process simulation
- Process optimization
- Product evaluation
- Optimal product
- Product + implicit application
- Product-process-application
-

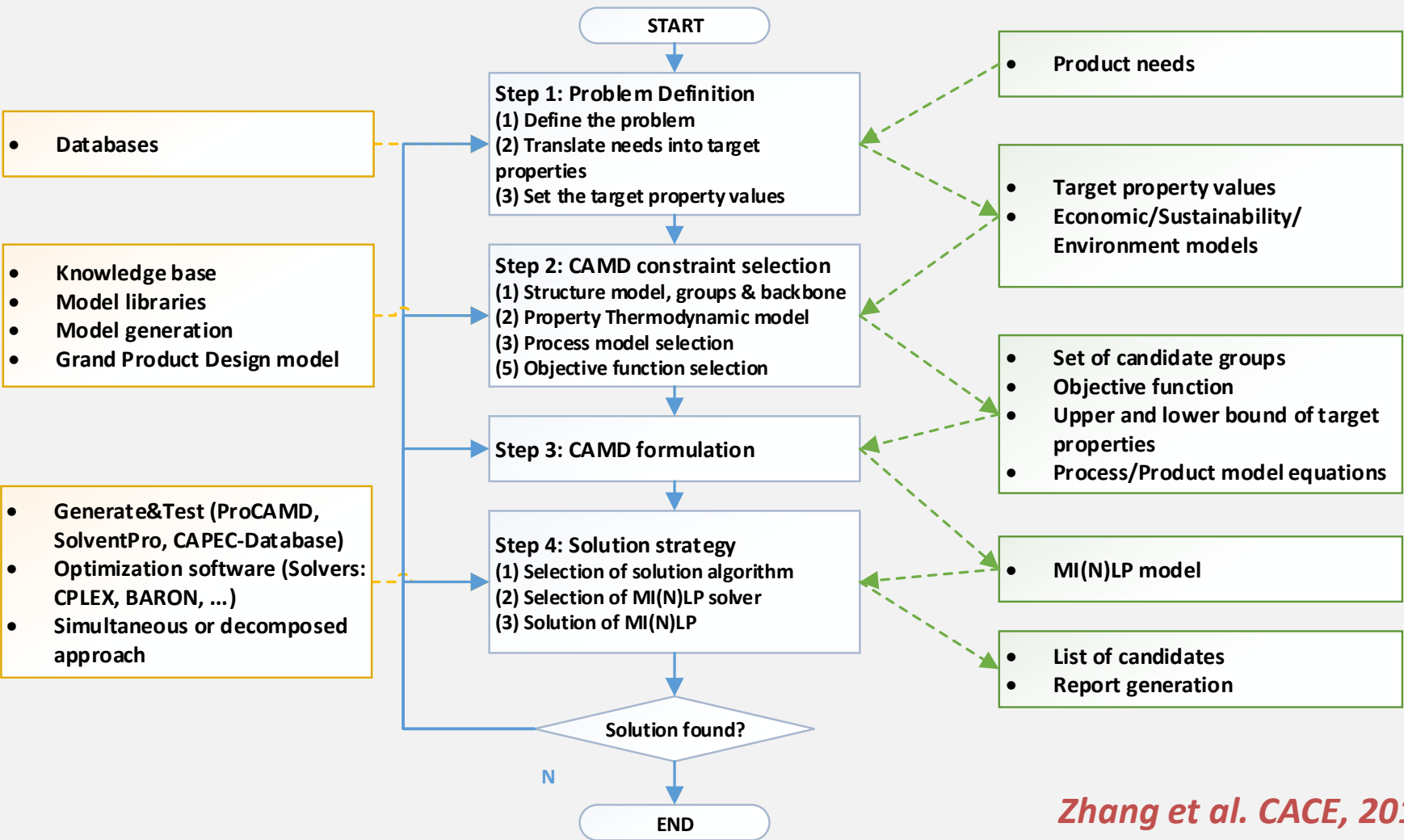


Simultaneous Product – Process Design (multiscale & multidiscipline)

Methods and Tools

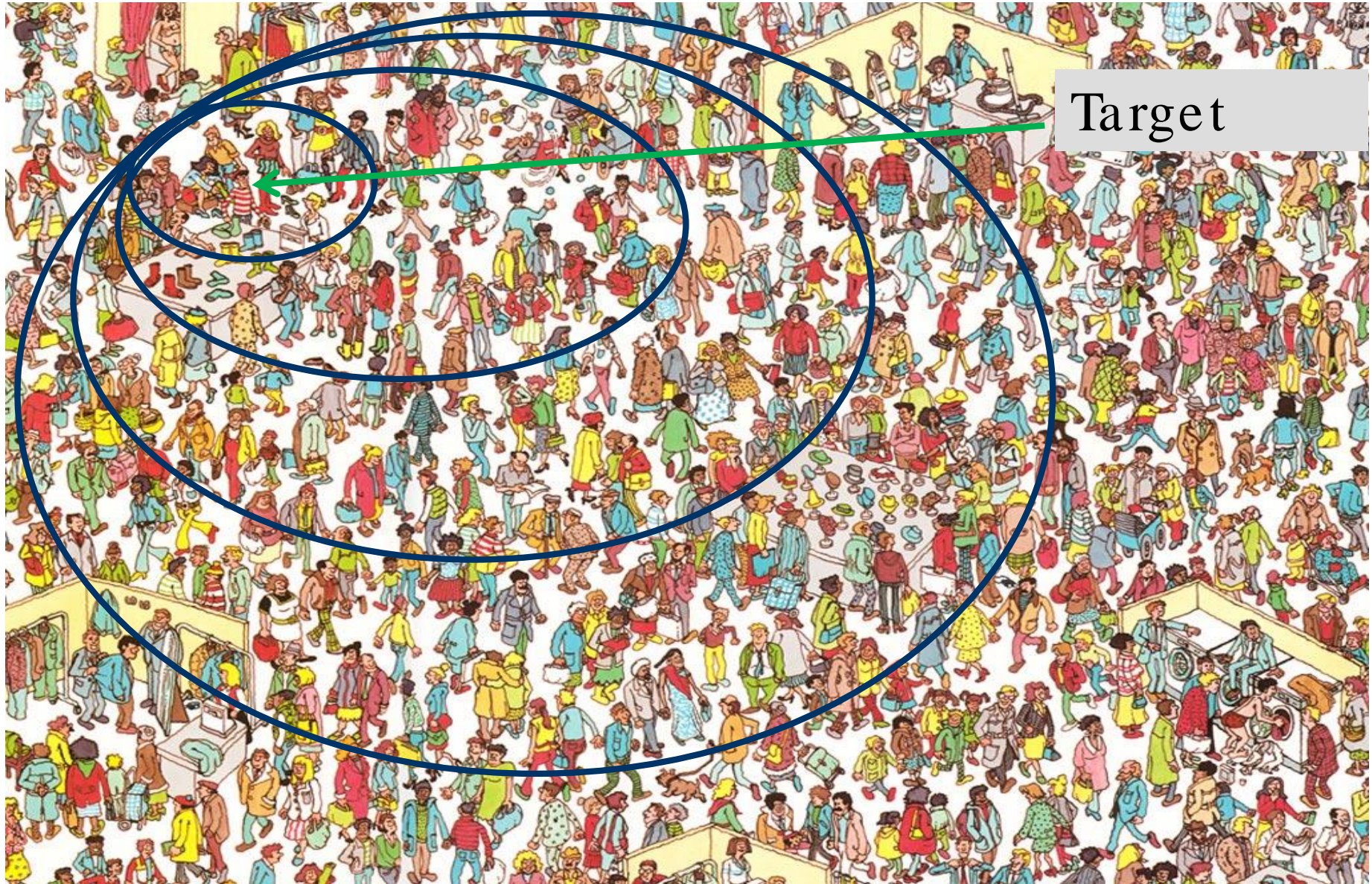
Work Flow

Information Flow

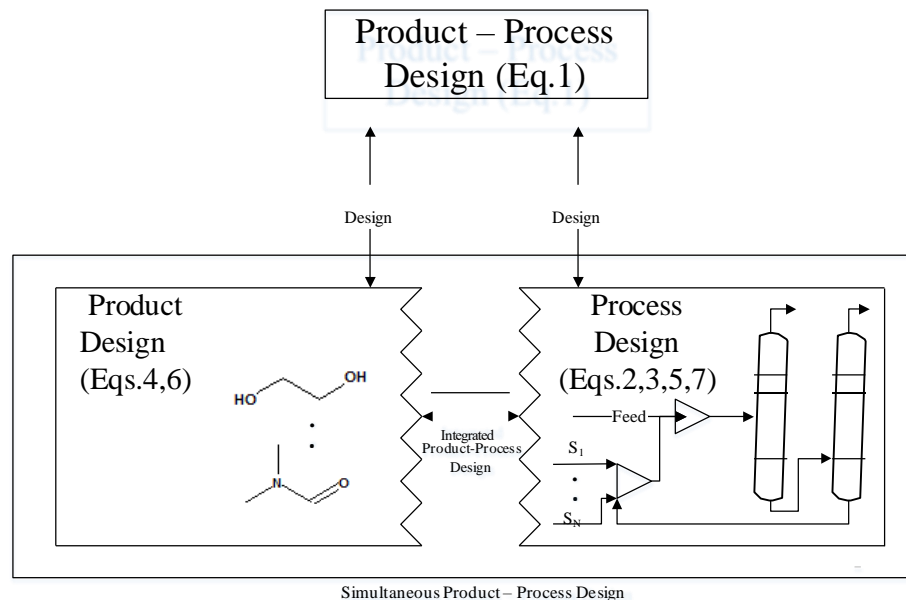


Zhang et al. CACE, 2015

Mathematical Problem Solution



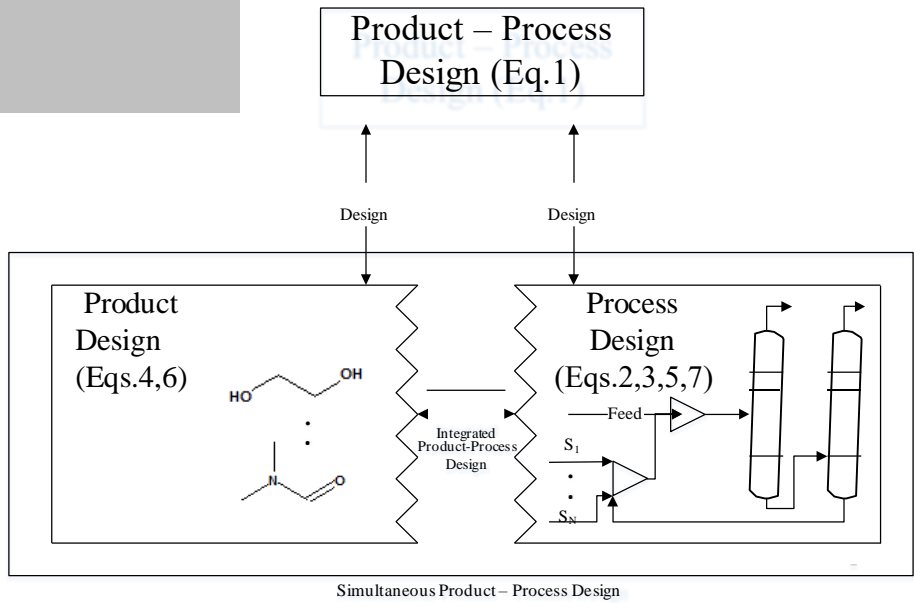
Target



**Eqs. 4, 6 with or without Eq. 1
(product design and implicit
performance verification)**

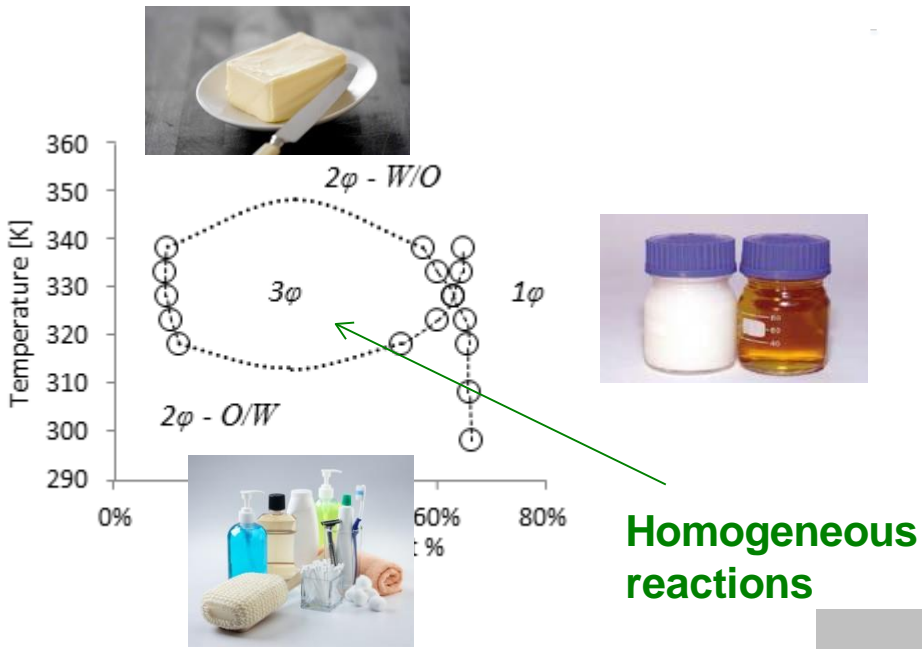
Examples: Molecular design (single species products)

- Solvents
- Process fluids
- **Surfactants**
- Active ingredients
- Polymer repeat units
-

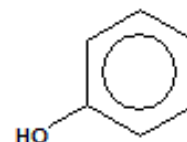


Eqs. 4, 6 with or without Eq. 1
(product design and implicit
performance verification)

Examples: Molecular
design (single molecular
products) - Surfactants



- **Step 1:** Design of a UV sunscreen, in the emulsified form, with a high sun protection factor. The phenol ring is fixed as backbone structure in this case study.
- **Step 2: Convert needs-functions to properties**
 - **Objective:** Toxicity should be minimized, but higher than a lower bound
 - **Group selection:** CH_3 , CH_2 , CH , C , aCH , aC-OH , CH_2COO , CH_3O , CH_2O , aC-O , $\text{OCH}_2\text{CH}_2\text{OH}$ from 220 groups (Marrero and Gani, (2001)).
 - **Backbone selection:** $-\text{C}_6\text{H}_4\text{OH}$



– **Structural and property constraints:**

| Need | Lower bound | Upper bound |
|-----------|--------------------------|----------------------|
| N_G | 10 | 15 |
| N_F | 0 | 8 |
| lc_{50} | $3.16\log(\text{mol/L})$ | - |
| Sp | - | $25\text{MPa}^{0.5}$ |
| clp | 343.15K | - |
| Fp | 343.15K | - |
| V_m | 0.1cc/mol | 0.3cc/mol |

LC_{50} : Fathead Minnow 96-hr exposure; Sp : Hildebrand solubility parameter; clp : cloud point; Fp : Flash point; V_m : Liquid molar volume at 298K.

• Step 3: Formulate the mathematical problem

$$\max lc50 = lco + \sum_{i_1} n_{i_1}^{(1)} lc50_{i_1}^{(1)} + \sum_{i_2} n_{i_2}^{(2)} lc50_{i_2}^{(2)}$$

2. Pure component property constraints:

1. Structural constraints:

1.1 Octet rule and group number constraints:

$$\sum_i (2 - \nu_{i_1}) n_{i_1}^{(1)} = 0$$

$$\sum_{i_1, i_1' \neq i_1'} n_{i_1}^{(1)} \geq n_{i_1'}^{(1)} (\nu_{i_1'} - 2) + 2 \quad \forall i_1'$$

$$1 \leq n_{i_1}^{(1)} \leq 5 \quad \forall i_1$$

$$10 \leq \sum_{i_1} n_{i_1}^{(1)} \leq 15$$

1.2 Adjacency matrix constraints.

1.3 Second-order group constraints.

$$lco + \sum_{i_1} n_{i_1}^{(1)} lc50_{i_1}^{(1)} + \sum_{i_2} n_{i_2}^{(2)} lc50_{i_2}^{(2)} \geq 3.16$$

$$Spo + \sum_{i_1} n_{i_1}^{(1)} Sp_{i_1}^{(1)} + \sum_{i_2} n_{i_2}^{(2)} Sp_{i_2}^{(2)} \leq 25$$

$$\sum_{i_1} n_{i_1}^{(1)} clp_{i_1}^{(1)} \geq 343.15^2$$

$$Fpo + \sum_{i_1} n_{i_1}^{(1)} Fp_{i_1}^{(1)} + \sum_{i_2} n_{i_2}^{(2)} Fp_{i_2}^{(2)} \geq 343.15$$

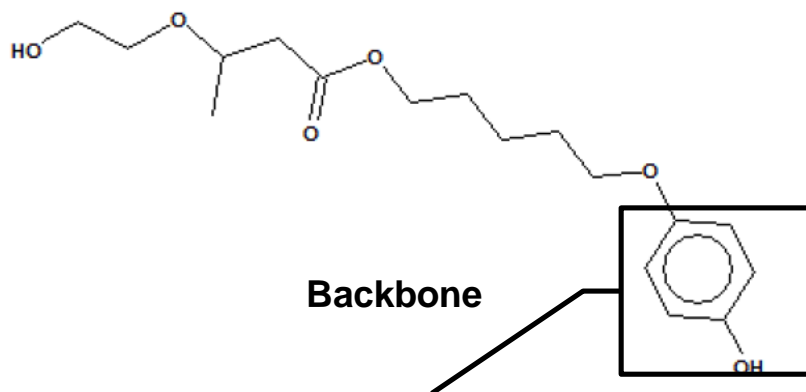
$$0.1 \leq Vmo + \sum_{i_1} n_{i_1}^{(1)} Vm_{i_1}^{(1)} + \sum_{i_2} n_{i_2}^{(2)} Vm_{i_2}^{(2)} \leq 0.3$$

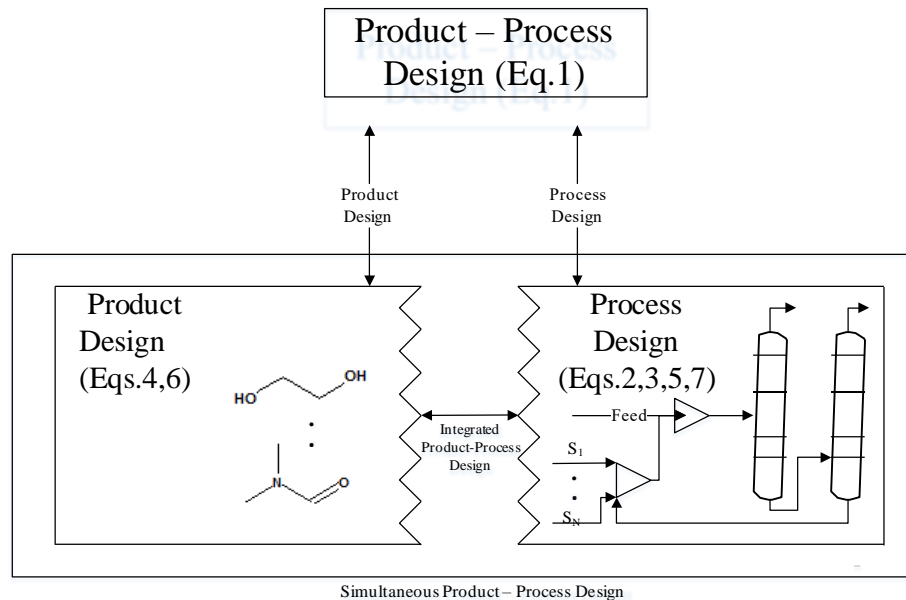
#Equations: 1,324,764; #Variables: 1,286,278 (1,286,266 Integer variables).

- **Step 4:** This design problem can also be solved directly using the **GAMS CPLEX** solver.

Table: Optimization results of surfactant design problem

| | |
|--------|--|
| n_1 | 1 CH ₃ , 5 CH ₂ , 1 CH, 4 aCH, 1 aC–OH, 1 CH ₂ COO, 1 aC–O, 1 OCH ₂ CH ₂ OH |
| n_2 | 1 AROMRINGS1s4 |
| $lc50$ | 4.124log(mol/L) |
| Sp | 24.732MPa ^{0.5} |
| clp | 300.016K |
| Fp | 552.383K |
| Vm | 0.281cc/mol |



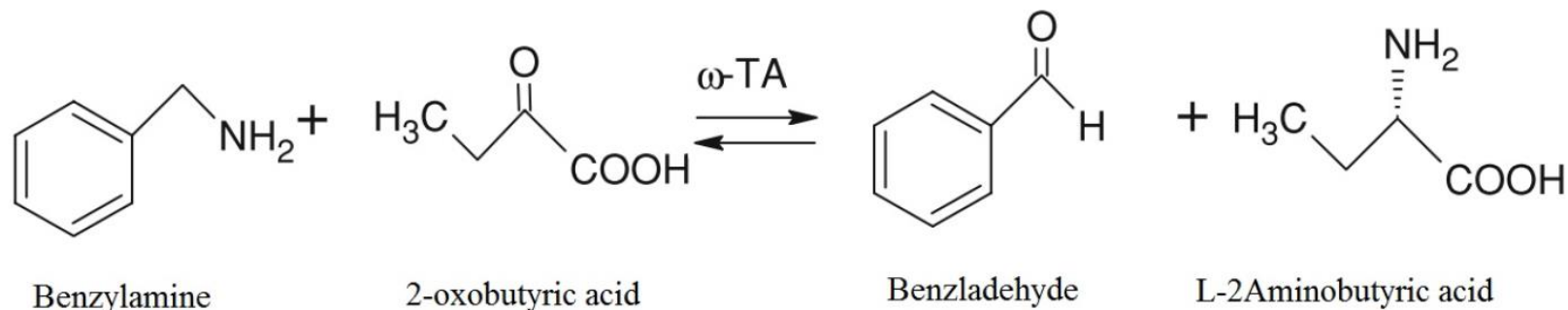


Examples: Molecular and/or mixture design with process application

- **Solvents for separations and/or product recovery**
- **Process fluids for specific operations**
- **Reactive agents**
-

Eqs. 2, 3, 4, 5, 6 with or without Eq. 1 (product-process development; solvent design-verification)

Generation of alternatives for the production of L-2-aminobutyric acid



Objective of the study: Identify possible process hotspots to improve the process in terms of:

- Conversion to continuous manufacturing
- Solvent selection to improve productivity
 - Reaction improvement
 - Separation improvement

SPEED Solvent based reaction-separation processes

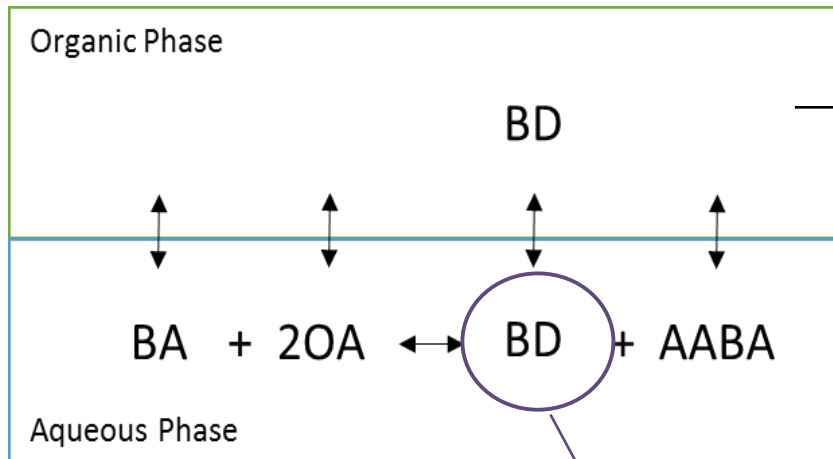
Generation of alternatives for the production of L-2-aminobutyric acid

Reaction type Database



- A. Single phase system
- B. Biphasic system

Experimental data available
Kinetic model is not available



Inhibits the enzyme

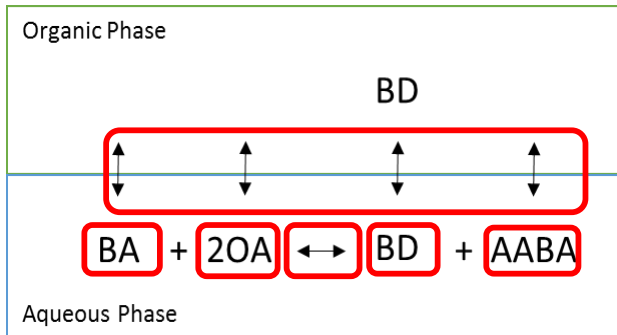
Solvent selection:

Organic solvent that is,

- Immiscible with the aqueous phase
- Higher by-product selectivity

SPEED Solvent based reaction-separation processes

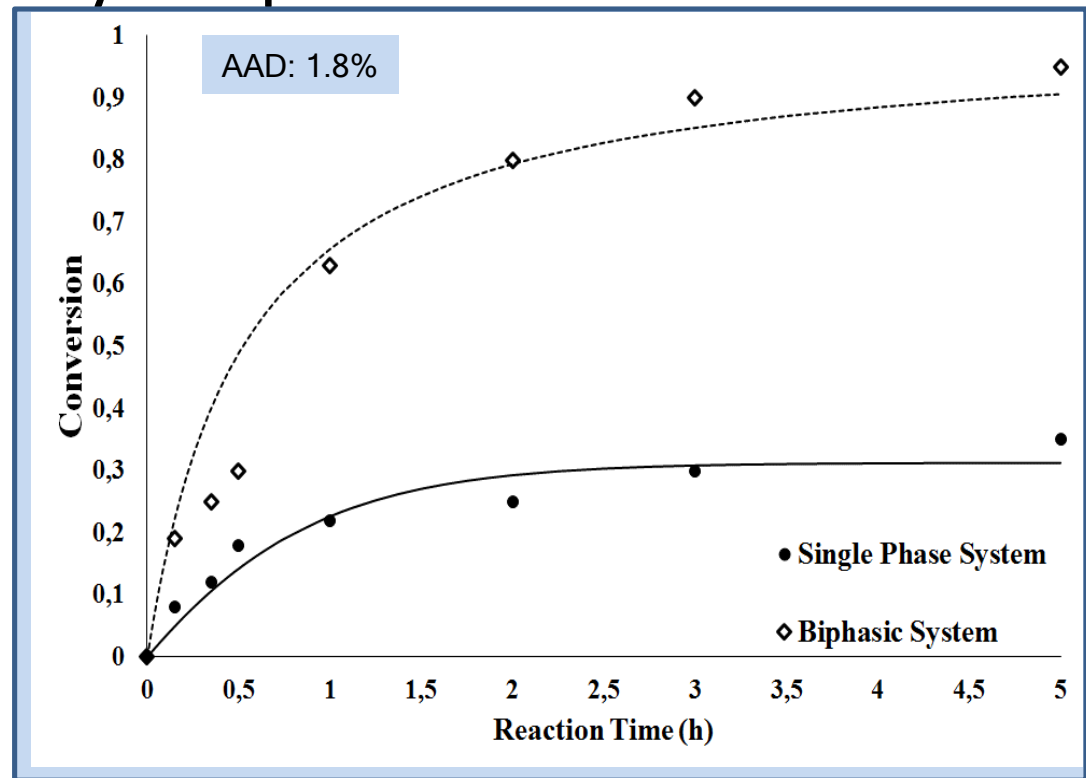
Generation of alternatives for the production of L-2-aminobutyric acid



Model development method: Multiphase reaction modeling (Anantpinijwatna et al. 2016)

Physical Equilibrium

Mass Balance



Reaction type Database:



Experimental data

SPEED Solvent based reaction-separation processes

Generation of alternatives for the production of L-2-aminobutyric acid

Extractive solvent selection:

| Partition $\left(\frac{\gamma_i^\alpha}{\gamma_i^\beta} \right)$ | Benzylamine | 2OA | AABA | Benzaldehyde |
|---|-------------|----------|----------|--------------|
| Hexane | 0.986 | 8.33E-06 | 4.50E-06 | 8.06E+03 |
| Heptane | 1.178 | 5.41E-06 | 3.56E-06 | 8.33E+03 |
| Isooctane | 3.096 | 8.33E-06 | 6.85E-06 | 1.59E+04 |
| Octane | 2.959 | 3.73E-04 | 1.19E-04 | 1.30E+04 |

Batch to continuous :



Batch to continuous
database

Enzyme membrane reactor

Facilitates enzyme stability and avoid the enzyme contact with the organic solvent in a continuous operation

SPEED Solvent based reaction-separation processes

Generation of alternatives for the production of L-2-aminobutyric acid

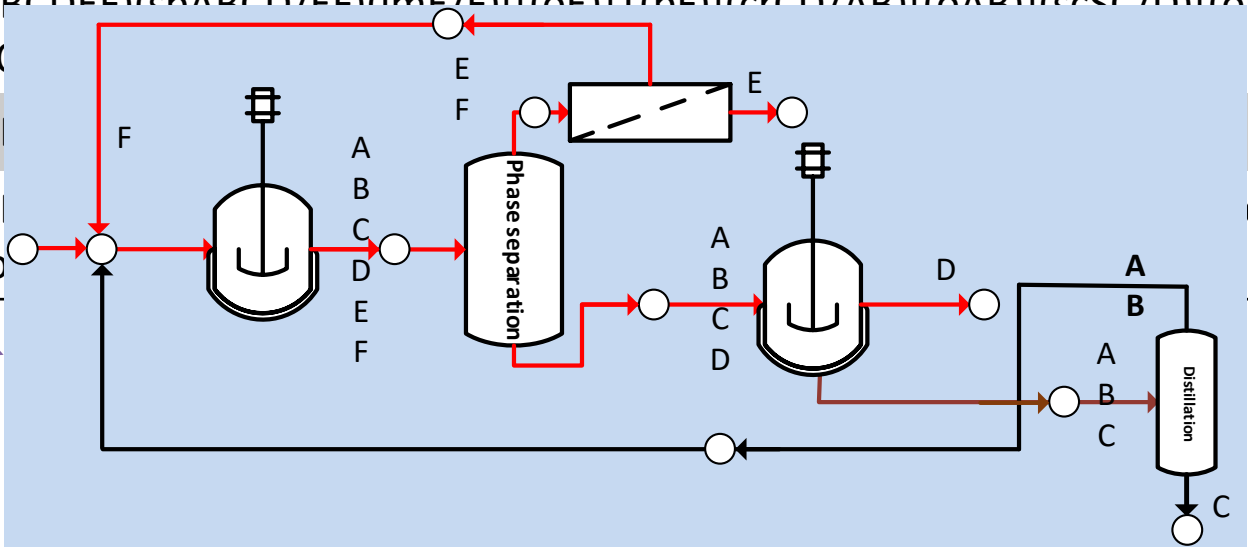
Objectives:

- Recover the main product
- Recover and recycle the extraction solvent (Hexane) and the reaction solvent (water)
- Recycle unreacted Benzamine

| Rank | Process alternatives [SFILES] |
|------|-------------------------------|
|------|-------------------------------|

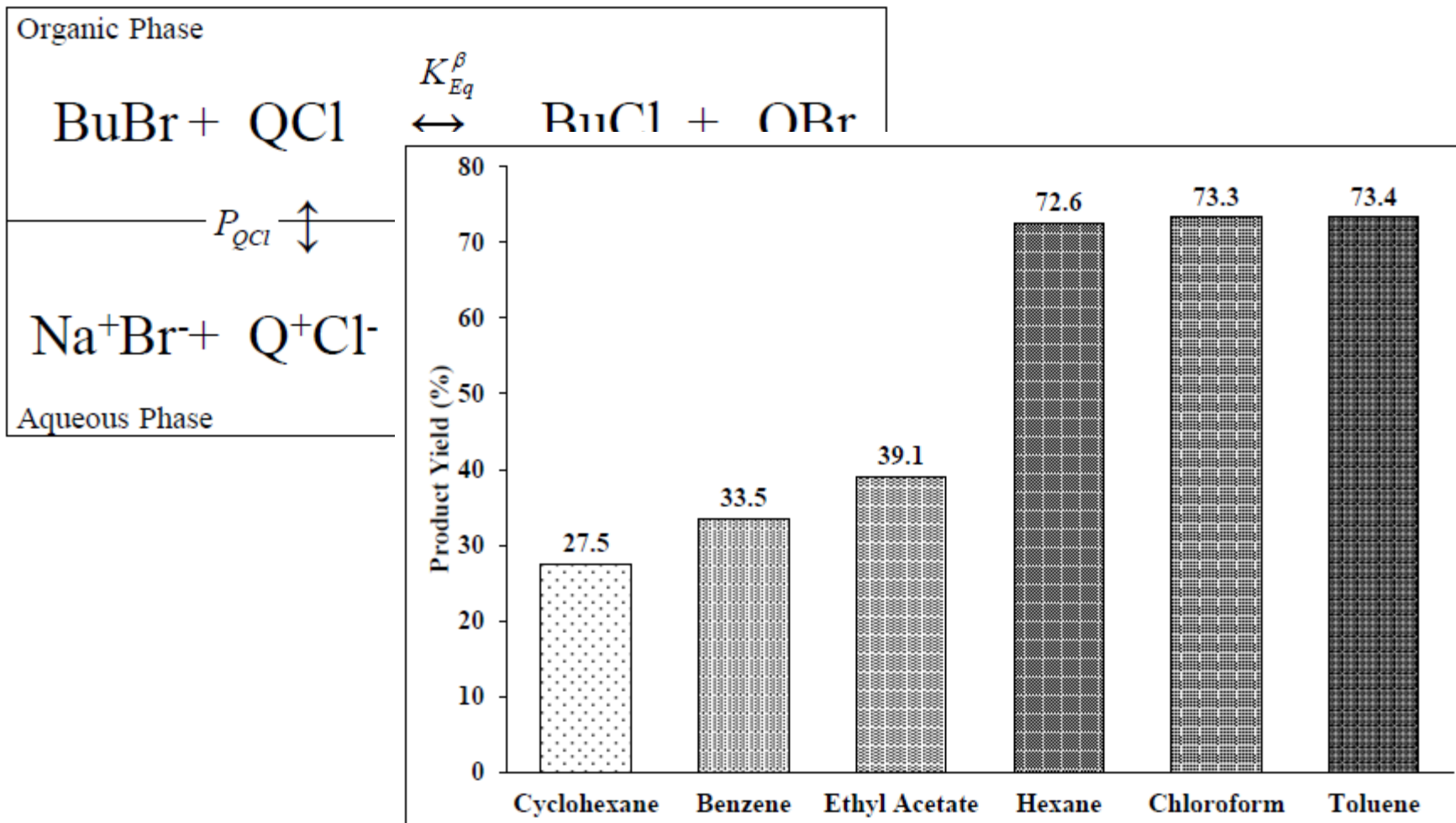
| | |
|---|---|
| 1 | (rABCDEF)(spABCD/EF)(lmE/F)[[(oE)]1(pF)](crD/ABC)[(oD)](dsAB/C)[(oAB)] (oC) |
| 2 | (rABCDEF)(spABCD/EF)(lmE/F)[[(oE)]1(pF)](crD/ABC)[(oD)](dsAB/C)[(oAB)] (oC) |

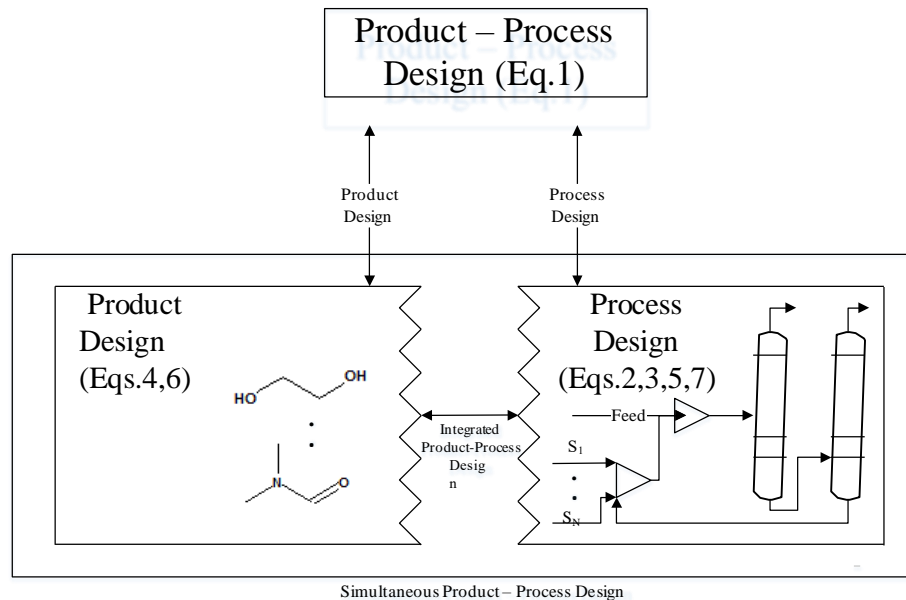
| | |
|---|---|
| 3 | (rABCDEF)(spABCD/EF)(lmE/F)[[(oE)]1(pF)](crD/ABC)[(oD)](dsAB/C)[(oAB)] (oC) |
| 4 | (rABCDEF)(spABCD/EF)(lmE/F)[[(oE)]1(pF)](crD/ABC)[(oD)](dsAB/C)[(oAB)] (oC) |



A: water
B: BA
C: 2OA
D: AABA
E: BD
F: Hexane

Other biphasic reaction-separation systems





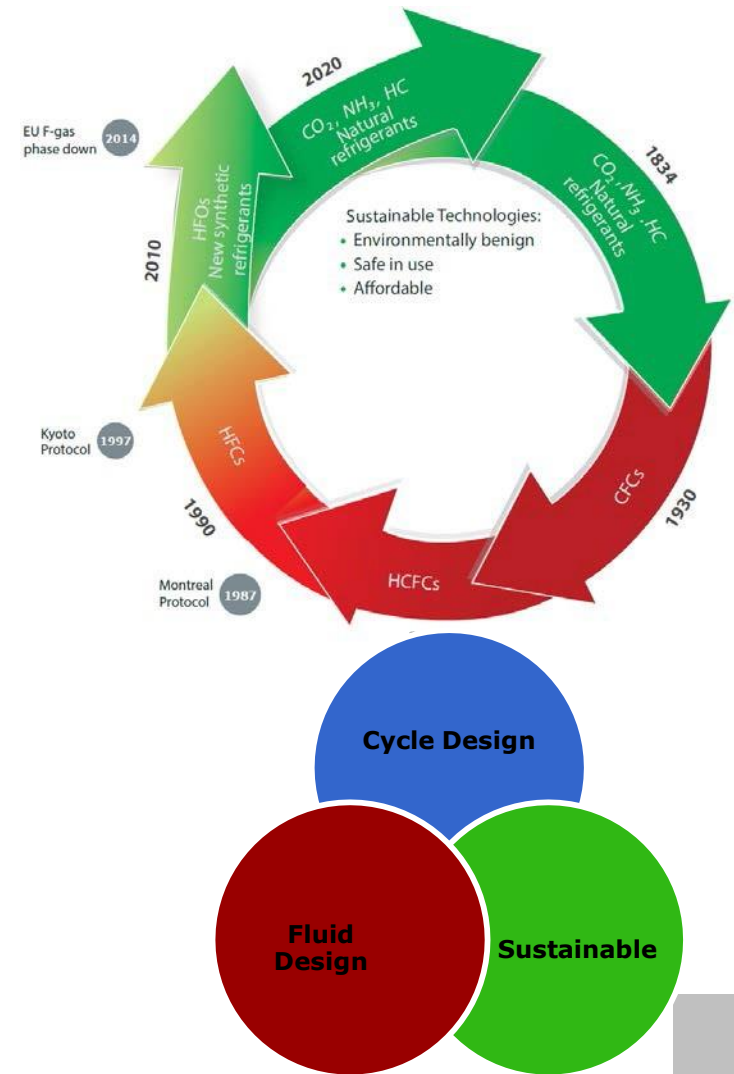
Eqs. 1-7: simultaneous product design & process application

Examples: Molecular design (single molecular products) with process constraints

- Solvents for separations and/or product recovery
- **Process fluids for specific operations**
- Reactive agents
-

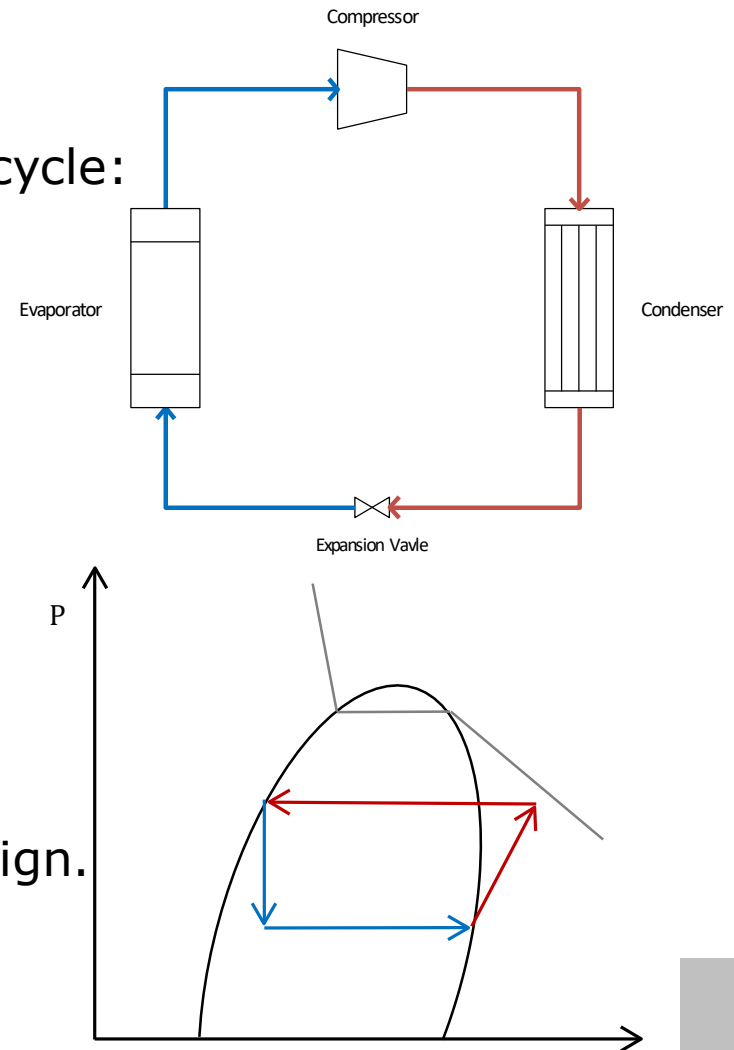
Refrigerant and cycle design-1

- Refrigeration cycle and fluid affect each other. Therefore, product and process application considered.
 - Objective is to determine novel working fluids and mixtures that comply with current and upcoming regulations.
 - Requirements for substitution of existing fluids:
 - New fluid should have same or improved cycle efficiency.
 - New fluid should have similar operational behavior.
 - New fluid should improve sustainability.
- Relate process with product properties.
- Find optimal product-cycle design.



Refrigerant-cycle design: Step 1

- R134a is a commonly used refrigerant that is being phased out.
- Product properties that affect the process cycle:
 - Critical properties;
 - Heat capacity;
 - Compressibility;
 - Accentric factor;
 - Vapor pressure;
 - Heat of vaporization...
- These product properties influence the process cycle and its efficiency.
- Also, other properties (enthalpy, entropy, fugacity) are needed for process cycle design.
- Equation of state and for estimation of PVT relationship.



SPEED

Simultaneous product and application process design

- Look for novel fluids that can contain C, H, F and Cl (up to 10 groups).
- 30 molecular groups are selected from M-G method.
- Target properties are chosen for product and process:

Step 2

| Target Property | Lower bound | Upper bound |
|-----------------------------------|-------------|-------------|
| Molecular mass | | 110 g/mol |
| Normal boiling point | | 250 K |
| Critical temperature | 350 K | 400 K |
| Critical pressure | 30 bar | 50 bar |
| Thermal conductivity (liquid, Tb) | 0.08 W/m.K | |
| Enthalpy of Vaporization at Tb | 200 kJ/kg | |
| ODP | | 0 |
| GWP | | 1400 |
| Atmospheric Lifetime * | | 14 years |
| Number of groups | 1 | 5 |
| Number of functional groups | 1 | 3 |
| Process needs | Lower bound | Upper bound |
| Evaporation temperature | 313.15 K | 333.15 |
| Condensation temperature | 265.65 K | 280.15 |

Refrigerant-cycle design: Step-3 (complete model-1)

- Molecular constraints

$$\sum_i n_i (2 - v_i) = 2$$

$$1 \leq \sum_i n_i \leq 10$$

$$0 \leq \sum_i n_i \leq 3$$

- Property constraints

$$\sum_i n_i M w_i \geq 110 \text{ g / mol}$$

$$T_{b0} \ln \left(\sum_i n_i T_{bi} \right) \geq 250 \text{ K}$$

$$350 \text{ K} \leq T_{c0} \ln \left(\sum_i n_i T_{ci} \right) \leq 400 \text{ K}$$

$$30 \text{ bar} \leq \frac{1}{\left(\sum_i n_i P_{ci} + P_{c02} \right)^2} + P_{c01} \leq 50 \text{ bar}$$

$$\frac{1.11}{\left(\sum_i n_i M w_i \right)^{1/2}} \times \frac{\left(3 + 20 \left(1 - \frac{T}{T_c} \right)^{2/3} \right)}{\left(3 + 20 \left(1 - \frac{T}{T_c} \right)^{2/3} \right)} \geq 0.1 \text{ W / m.K}$$

$$\Delta H_v(T_0) \left(\frac{1 - \frac{T_1}{T_c}}{1 - \frac{T_0}{T_c}} \right)^{0.38} \geq 200 \text{ kJ / kg}$$

- Process constraints

$$f(Z_{V1}) = 0 = Z_{V1}^3 - Z_{V1}^2 + Z_{V1}(-B^2 + A - B) - AB$$

$$f(Z_{V2}) = 0 = Z_{V2}^3 - Z_{V2}^2 + Z_{V2}(-B^2 + A - B) -$$

$$\frac{AB}{f(Z_{L3})} = 0 = Z_{L3}^3 - Z_{L3}^2 + Z_{L3}(-B^2 + A - B) - AB$$

Needed for efficient solution of cubic EoS [4]:

$$f'(Z_{V2}) \geq 0$$

$$f(Z_{V1}) = 0 \quad f''(Z_{V2}) \geq 0$$

$$f'(Z_{V1}) \geq 0 \quad f(Z_{L3}) = 0$$

$$f''(Z_V) \geq 0 \quad f'(Z_{L3}) \geq 0$$

$$f(Z_{V2}^1) = 0 \quad f''(Z_{L3}) \leq 0$$

4] Kamath, R.S., Biegler, L.T., Grossmann, I.E., 2010, Comput. Chem. Eng. 34, 2085–2096.

Refrigerant-cycle design: Step-3 (complete model-2)

• Process constraints

$$\ln \phi_{V1} = \ln \phi_{L1}$$

$$\ln \phi_{V2} = \ln \phi_{L2}$$

$$H_{L3}(T_3, P_3) - H_{L4}(T_4, P_4) = 0$$

$$S_{V2}(T_2, P_2) - S_{V1}(T_1, P_1) = 0$$

$$P_1 = P_4$$

$$P_2 = P_3$$

$$T_1 = T_4$$

$$P_1 \geq P_0$$

$$P_2 \leq 0.8P_c$$

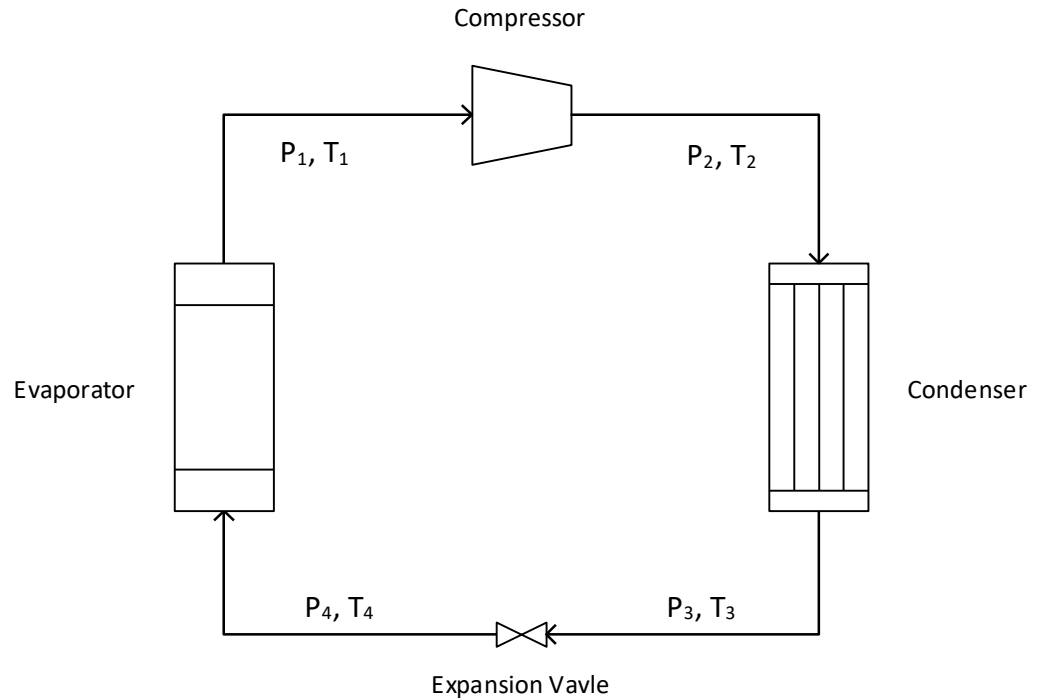
$$T_2 \leq 0.8T_c$$

$$265.65K \leq T_1 \leq 280.15K$$

$$313.15K \leq T_3 \leq 333.15K$$

• Objective function

$$\max COP = \frac{H_{1, V1}(Z_{V1}, T_1, P_1) - H_{3, L3}(Z_{L3}, T_3, P_3)}{H_{2, V2}(Z_{V2}, T_2, P_2) - H_{1, V1}(Z_{V1}, T_1, P_1)}$$



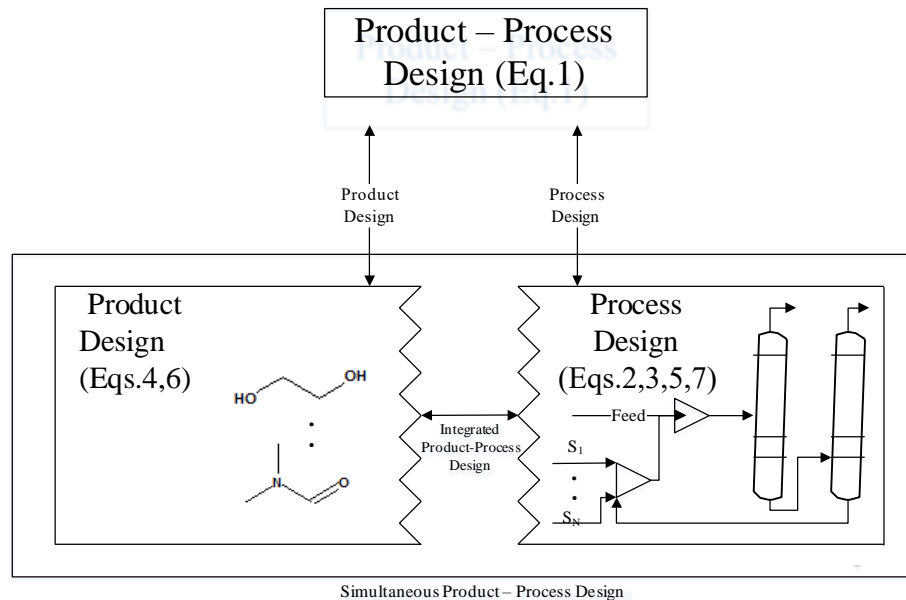
SPEED

Case study: Product-process design

Cycle design for each molecules is given by Problem (II)

Results

| | | r134a | r152a | 3,3,3-triFluoro propyne | 3,3-diFluoro butene | 2,2-diFluoro butane |
|------------|-----------------------------------|------------------------------|-----------------------------|-------------------------|---------------------|---------------------|
| Problem I | Mw | 102 g/mol | 66 g/mol | 94 g/mol | 92 g/mol | 94.1 g/mol |
| | Tb | 236 K | 230 K | 242 K | 246 K | 252 K |
| | Tc | 374 K | 366 K | 369 K | 392 K | 389 K |
| | Pc | 37.32 bar | 46.75 bar | 38.13 bar | 39.27 bar | 37.8 bar |
| | ω | 0.33 | 0.27 | 0.28 | 0.14 | 0.18 |
| | k | 0.082 W/m.k | 0.097 W/m.k | 0.086 W/m.K | 0.093 W/m.K | 0.092 W/m.K |
| | ΔH _v at T _b | 26523 J/mol | 28562 J/mol | 18635 J/mol | 21312 J/mol | 26991 J/mol |
| | ODP | 0 | 0 | 0 | 0 | 0 |
| | GWP | 1370 (ASHRAE, 2005) | 133 (ASHRAE, 2005) | 134.6 | 2.4 | 0.9 |
| | Atmospheric lifetime | 13.4 years (ASHRAE, 2005) | 1.5 years (ASHRAE, 2005) | - | - | |
| Problem II | COP | 10.1 | 12.8 | 9.5 | 11.0 | 11.1 |



Examples: Formulated and/or blend design

- Liquid formulated products
- Tailor-made blends
- Emulsified products
-

Eqs. 1, 4 & 6: mixture-blend design

SPEED Tailor-made fuel blends - 1

The main challenge involves how to identify the blends that satisfy the blend target properties with various types of additives

Problems solved: Tailor made design of gasoline blends, diesel blends, jet-fuel blends and lubricant blends.

Example: Blending gasoline with different **additives:**

- ✓ impact of fuel consumption on the environment.
- ✓ performances can be retained and improved.

Blend templates help to efficiently narrow down the search space of feasible chemicals to be added and obtain improved tailor-made blends in a faster, reliable and robust manner



* Collaboration with Texas A&M in Qatar.

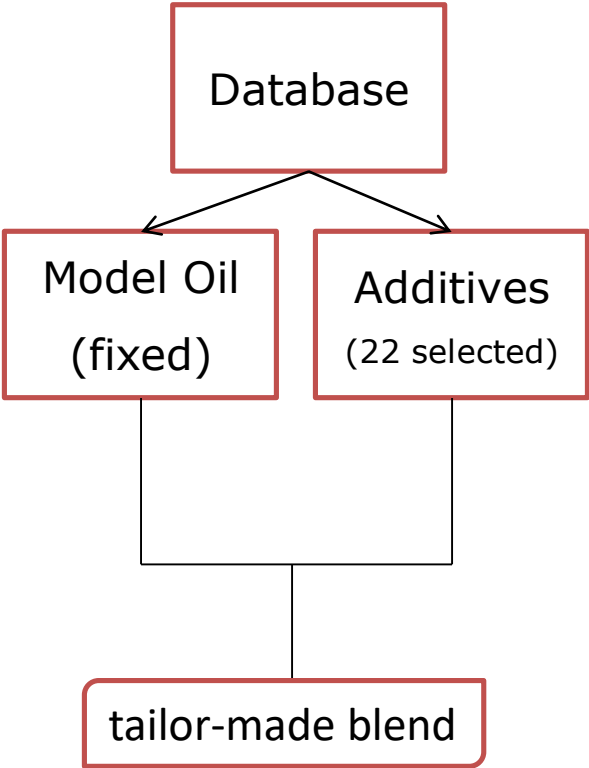
**** To be presented at Session I-2,
paper OL5, 16:40 by Sawitree Kalakul***

SPEED Tailor-made fuels - 2

Task 1 Problem Definition

- 1.1 Identify product needs
- 1.2 Translate needs into physico-chemical properties
- 1.3 Set target values

| Need | Target property | Target value |
|--------------------------|---------------------|------------------------------|
| Ability to be burned | RVP | $45 \leq RVP \leq 60$ |
| Engine efficiency | RON | $RON \geq 92$ |
| | HHV | $HHV \geq 40$ |
| Consistency of fuel flow | η | $0.30 \leq \eta \leq 0.60$ |
| | ρ | $0.720 \leq \rho \leq 0.775$ |
| Flammability | T_f | $T_f \leq 300$ |
| Toxicity | LC_{50} | $-\log LC_{50} < 3.08$ |
| Stability | ΔG^{mix} | $\Delta G^{mix} < 0$ |
| Environmental aspect | Wt_{O_2} | $2 \leq Wt_{O_2} \leq 20$ |
| Low oxidation | Choice of chemicals | |



SPEED Tailor-made fuels

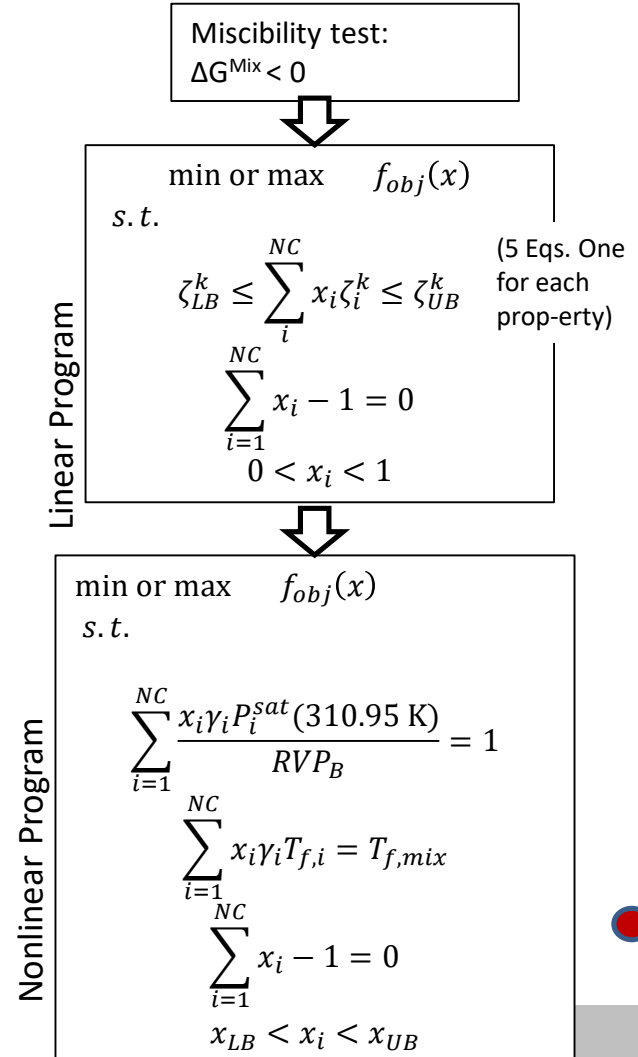
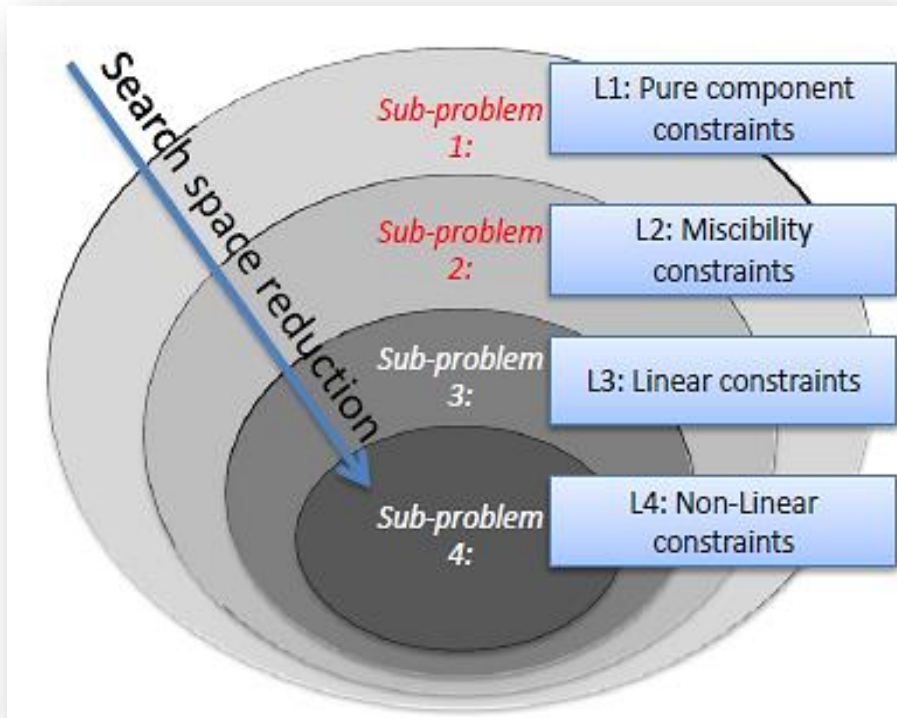
Task 2 Mixture/blend design

2.1 Pure component constraints (HHV, ρ , η , $LC50$, $WtO2$)

2.2 Miscibility constraint (ΔG^{mix})

2.3 Linear constraints (HHV, ρ , η , $LC50$, $WtO2$)

2.4 Non-linear constraints (RVP, T_f)

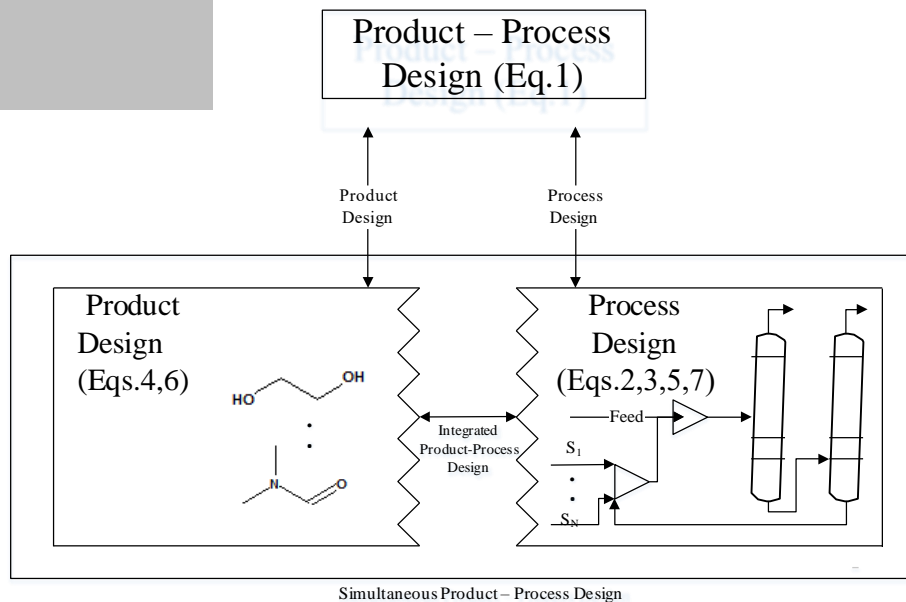


Task 3 Rank blend candidates according to a selection criterion

| Blend | Composition (vol%) |
|---------|--|
| Blend 1 | MI (69), Tetrahydrofuran (11), 2-Methyl Tetrahydrofuran (20) |
| Blend 2 | MI (67), Acetone (13), 2-Methyl Tetrahydrofuran (20) |
| Blend 3 | MI (72), Acetone (10), 2-Butanone (18) |
| Blend 4 | MI (75), 2-Butanone (13), 2-Methyl Tetrahydrofuran (12) |
| Blend 5 | MI (77), Ethanol (12), 2-Methyl Tetrahydrofuran (11) |

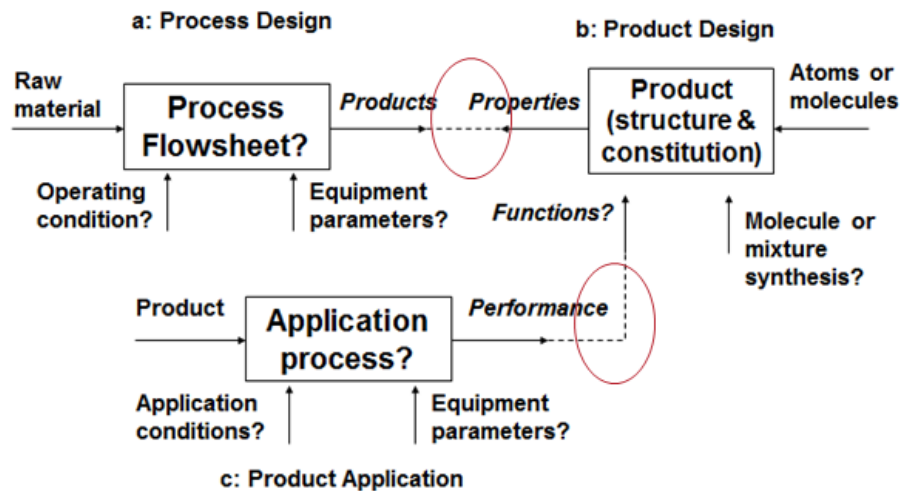
Task 4 Experimental verification

| Sample | Density at 15°C (g/[cm] ³) | | Dynamic Viscosity(mPa.s) | | Vapor Pressure at 37.8°C (kPa) | |
|-----------------|--|-------|--------------------------|-------|--------------------------------|-------|
| Target Values | 0.720 – 0.775 | | 0.3 – 0.6 | | 45 - 60 | |
| | Experimental | Model | Experimental | Model | Experimental | Model |
| Main Ingredient | 0.71 | 0.73 | 0.50 | 0.51 | 51.0 | 54.0 |
| Blend 1 | 0.76 | 0.76 | 0.54 | 0.46 | 46.2 | 50.8 |
| Blend 2 | 0.75 | 0.74 | 0.46 | 0.43 | 60.4 | 64.5 |
| Blend 3 | 0.73 | 0.74 | 0.45 | 0.43 | 58.9 | 63.7 |
| Blend 4 | 0.74 | 0.73 | 0.46 | 0.45 | 50.1 | 53.5 |
| Blend 5 | 0.74 | 0.74 | 0.61 | 0.58 | 55.1 | 58.9 |



Some interesting developments:

- **The chemical product simulator**
- **Product-process development**

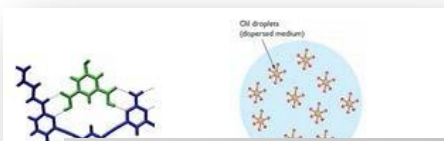




There is a need for a product simulator with similar and more useful features than a typical process simulator.

Based on available data, models, methods and analysis tools, the first chemical product simulator has been developed: **ProCAPD**

Kalakul et al, "VPPD-Lab: The Chemical Product Simulator", in "Methods and tools for chemical product design", Computer Aided Chemical Engineering, 39, 61-94 (chapter 3), 2017



Product design

Use design templates
(molecule products, formulated products, blended products, emulsified products and devices)



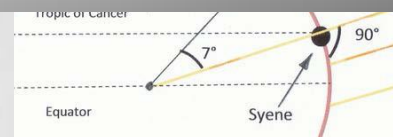
Product performance

Simulated product performance through virtual application experiments



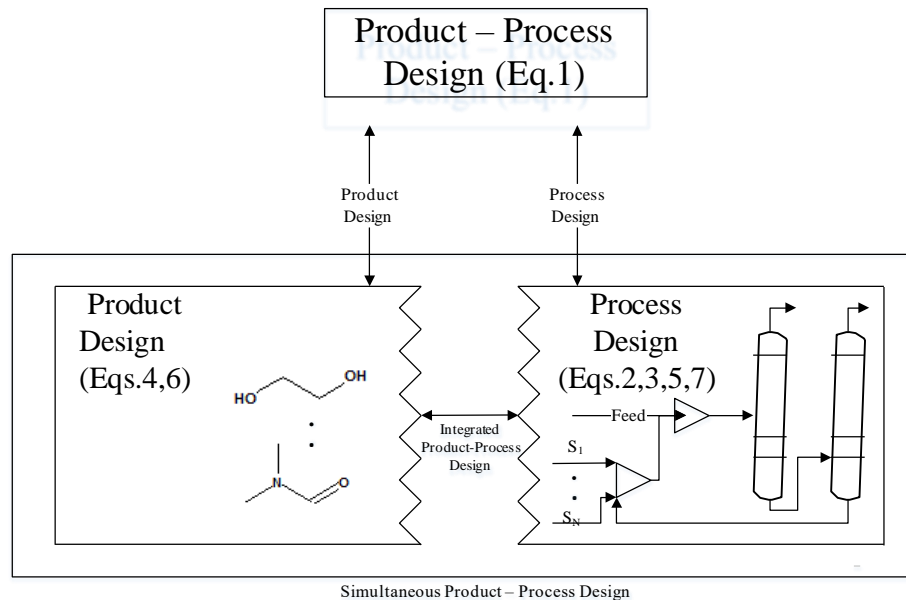
Product analysis

Predict and analyze product behavior
(identify important product properties)



Product search

Search for data, models, properties, products, devices, etc.

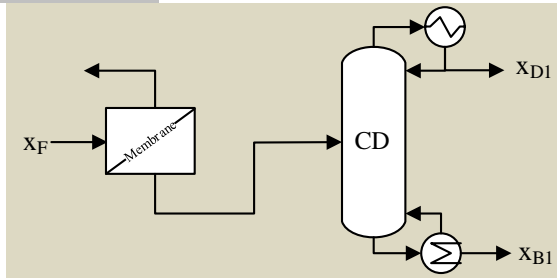


Eqs. 1-7: simultaneous product design & process application – decomposition-based solution approach

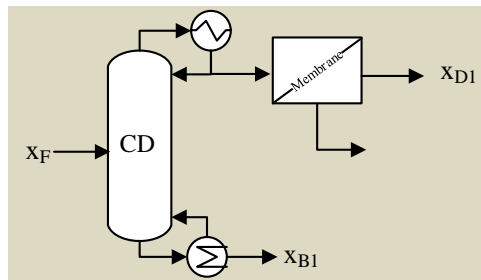
Examples: Molecular design (single molecular products) with process constraints

- Solvents for separations and/or product recovery
- **Hybrid process design with membranes**
- Process fluids for specific operations
- Reactive agents
-

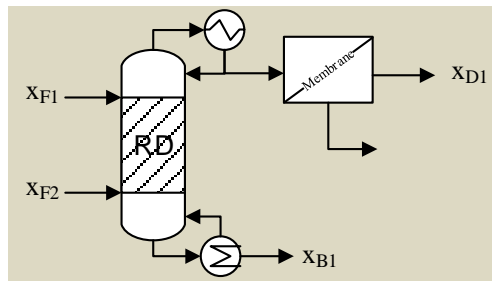
SPEED Design of hybrid modules



1

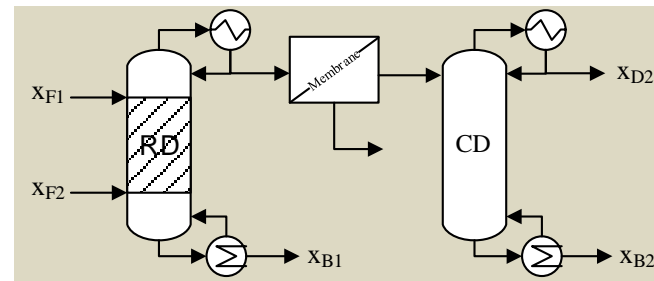
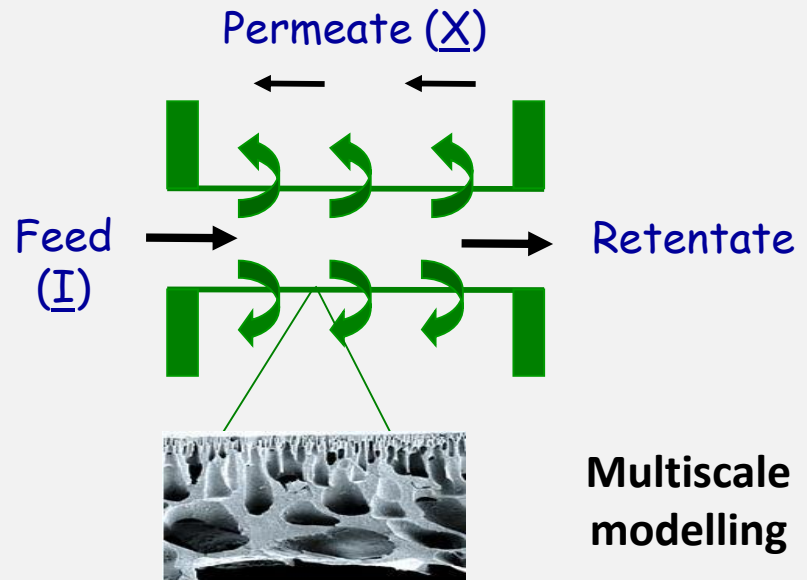


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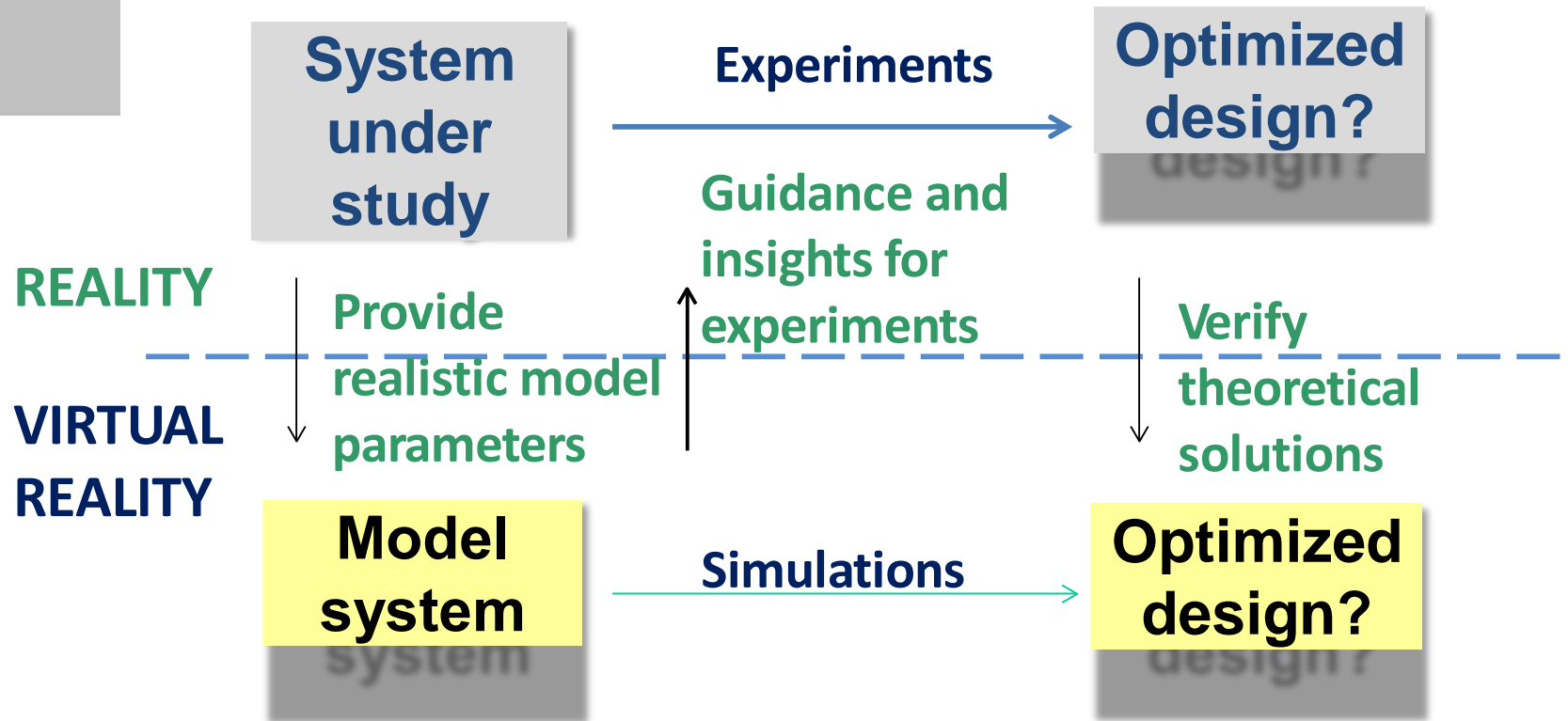
3

Which is the product design problem?

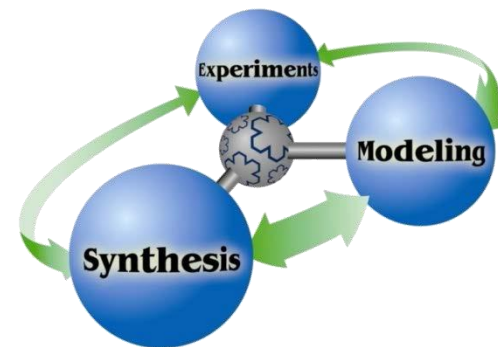


4

Close to 50% or more energy reduction compared to original process achievable

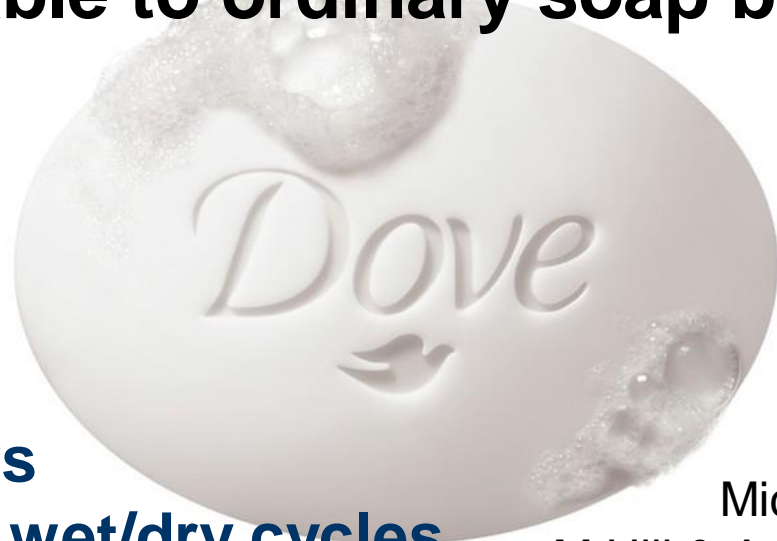


- Solution approaches
 - Integrated multiscale modeling, experiments and synthesis
 - Ability to find predictive-innovative solutions



Design Problem from 60 Years Ago

- **Non-scumming “soap bar”** (world’s best-selling soap bar) – Launched by Lever Bros (US division of Unilever) in 1955
- **Attributes comparable to ordinary soap bars**
 - **Firmness**
 - **Lather**
 - **Rate of wear**
 - **Slipperiness**
 - **Mildness**
 - **No unpleasant odors**
 - **No “cracking” from wet/dry cycles**
- **Processable on ordinary soap bar line: model-based system yet to be developed**

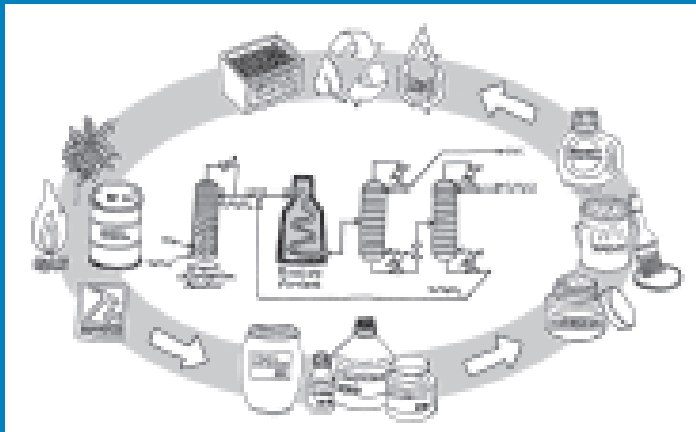


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